

Engineering the tryptophan synthase β -subunit for synthesis of non- canonical amino acids



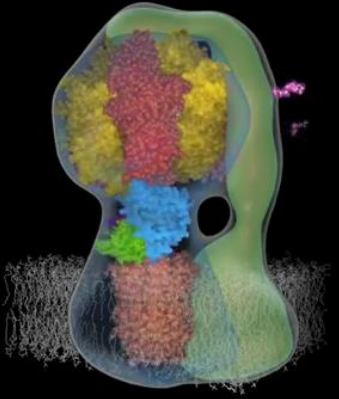
Ella Watkins-Dulaney

Thesis defense

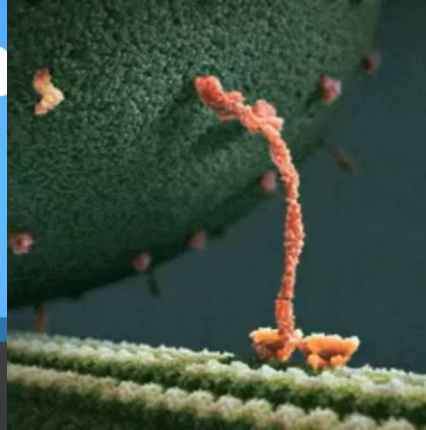
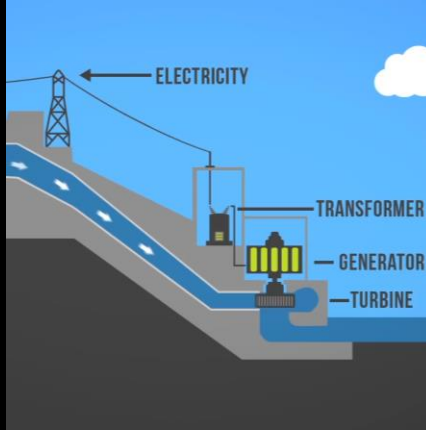
210610



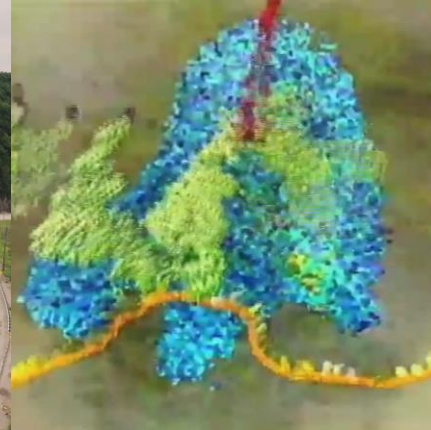
Proteins are incredible molecular machines



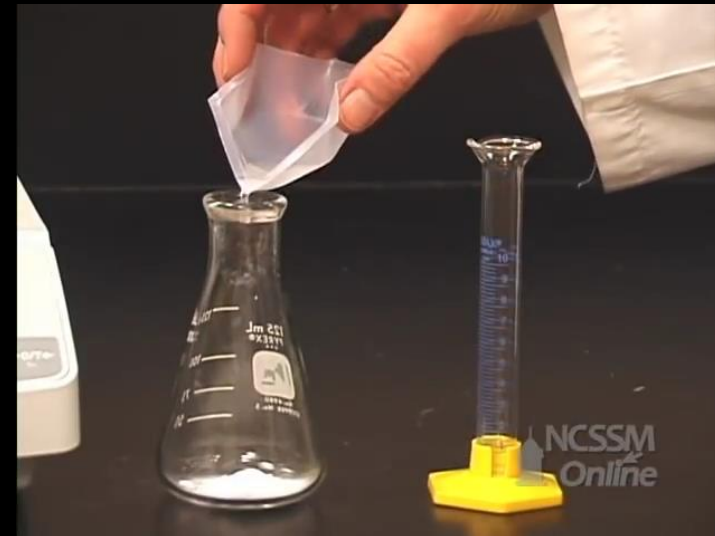
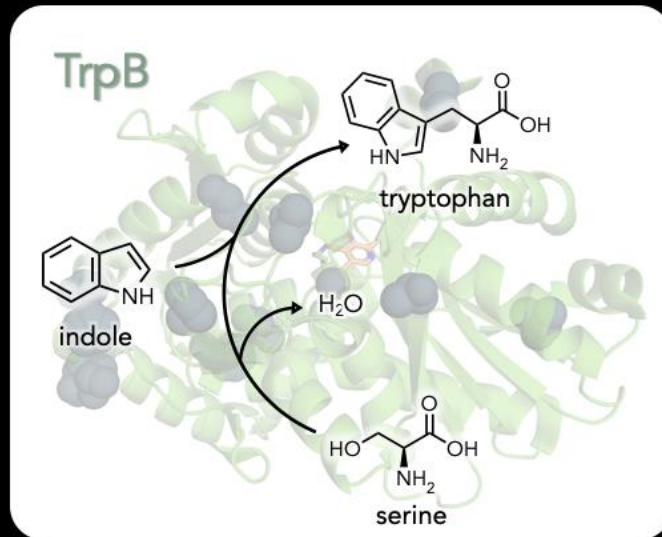
ATP synthase works just like a hydroelectric dam



Kinesin shuttles cargo across the cell



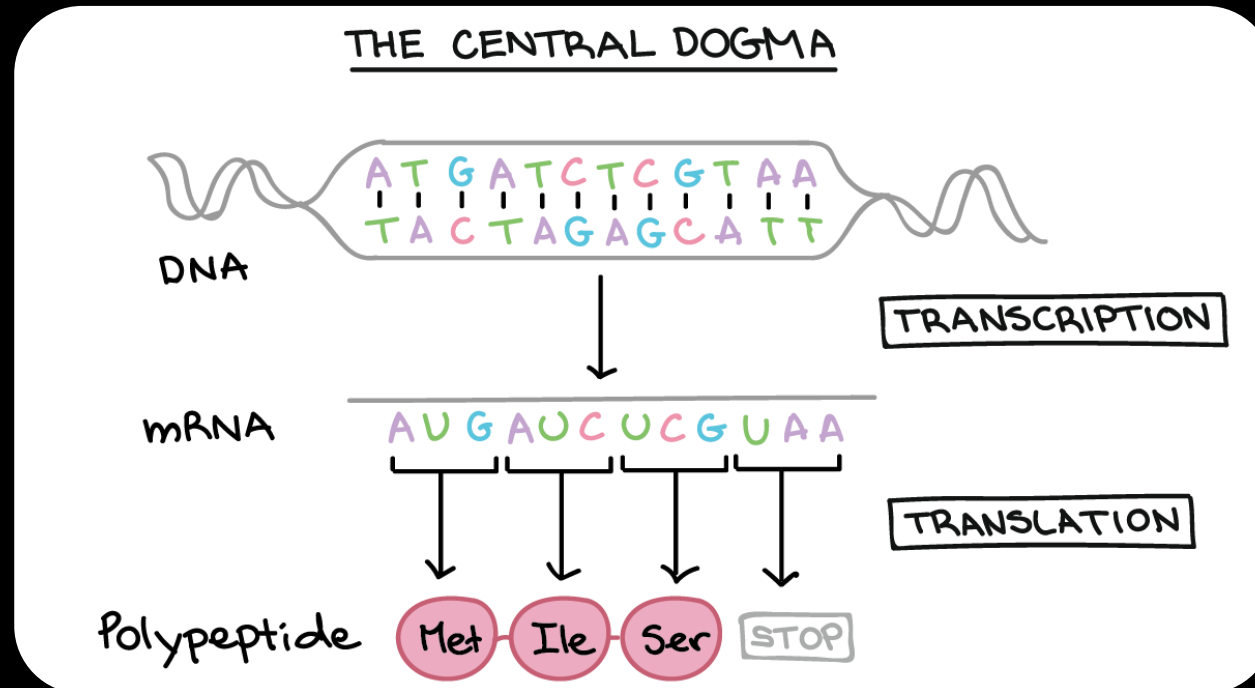
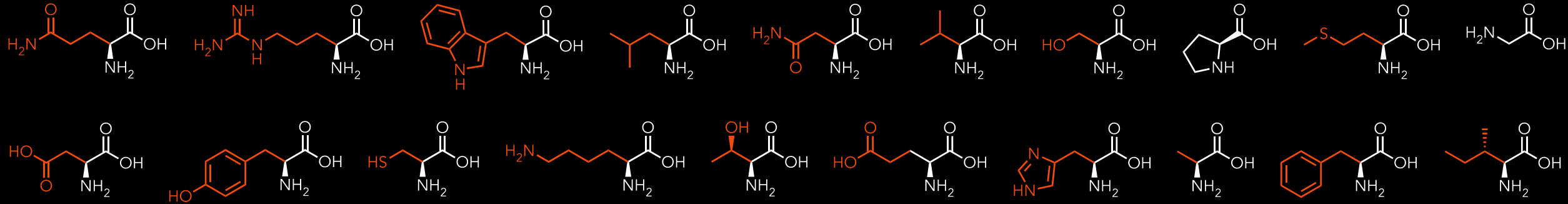
The ribosome is like an assembly line for protein synthesis



Enzymes are proteins that catalyze chemical reactions

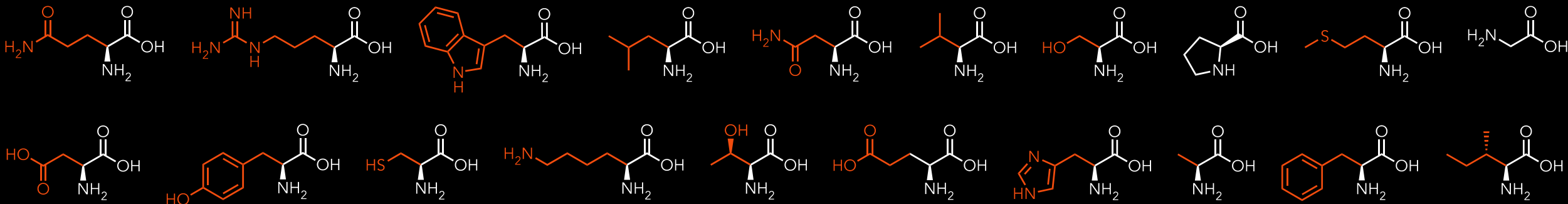
Proteins are made of 20 canonical amino acids

Canonical amino acids



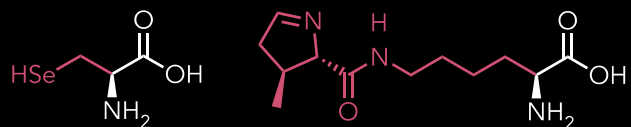
To infinity and beyond: non-canonical amino acids (ncAAs)

Canonical amino acids

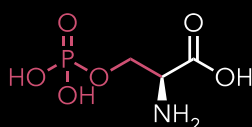


Naturally occurring non-canonical amino acids and modifications

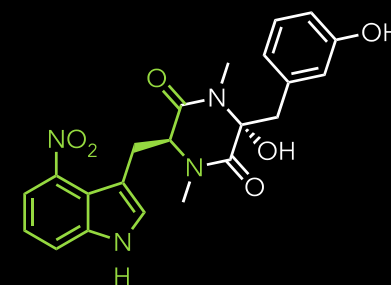
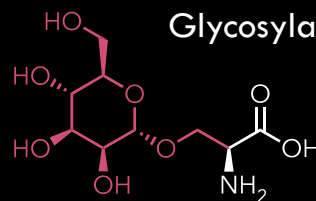
Genetic code expansion



Phosphorylation



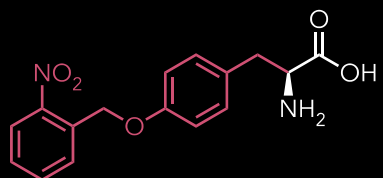
Glycosylation



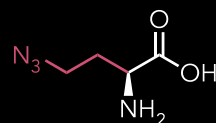
Thaxtomin A
(herbicide)

Unnatural non-canonical amino acids

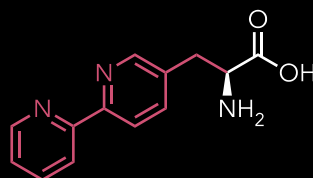
Photocaged



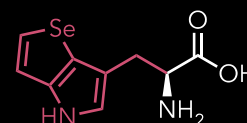
Bioorthogonal chemistry



Metal binding



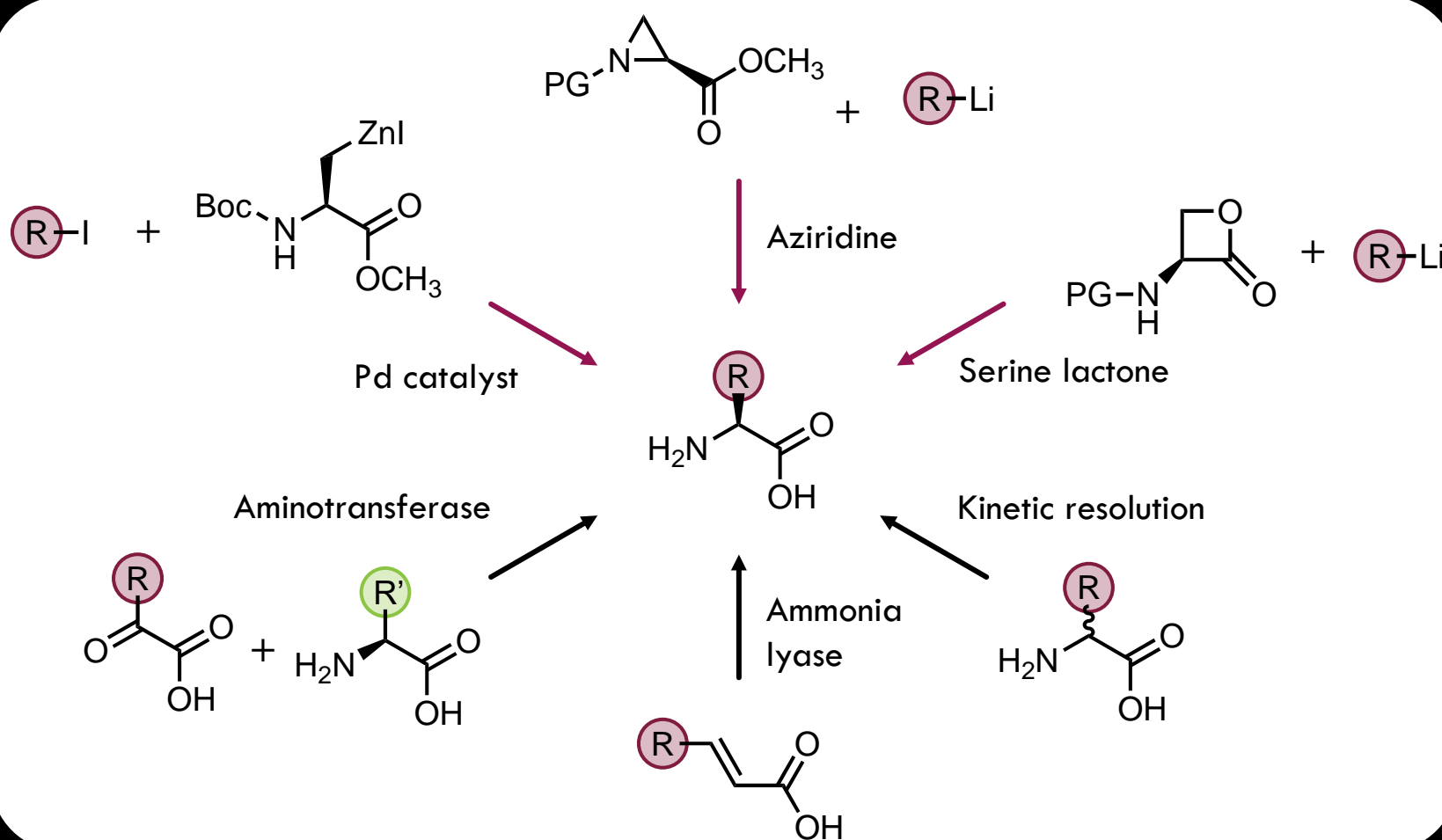
X-ray crystallography



Daptomycin
(antibiotic)

How do we access non-canonical amino acids?

Challenges with ncAA synthesis



Chemical synthesis

- ✗ Requires protection
- ✗ Not selective
- ✗ Catalyst synthesis
- ✓ Broad scopes

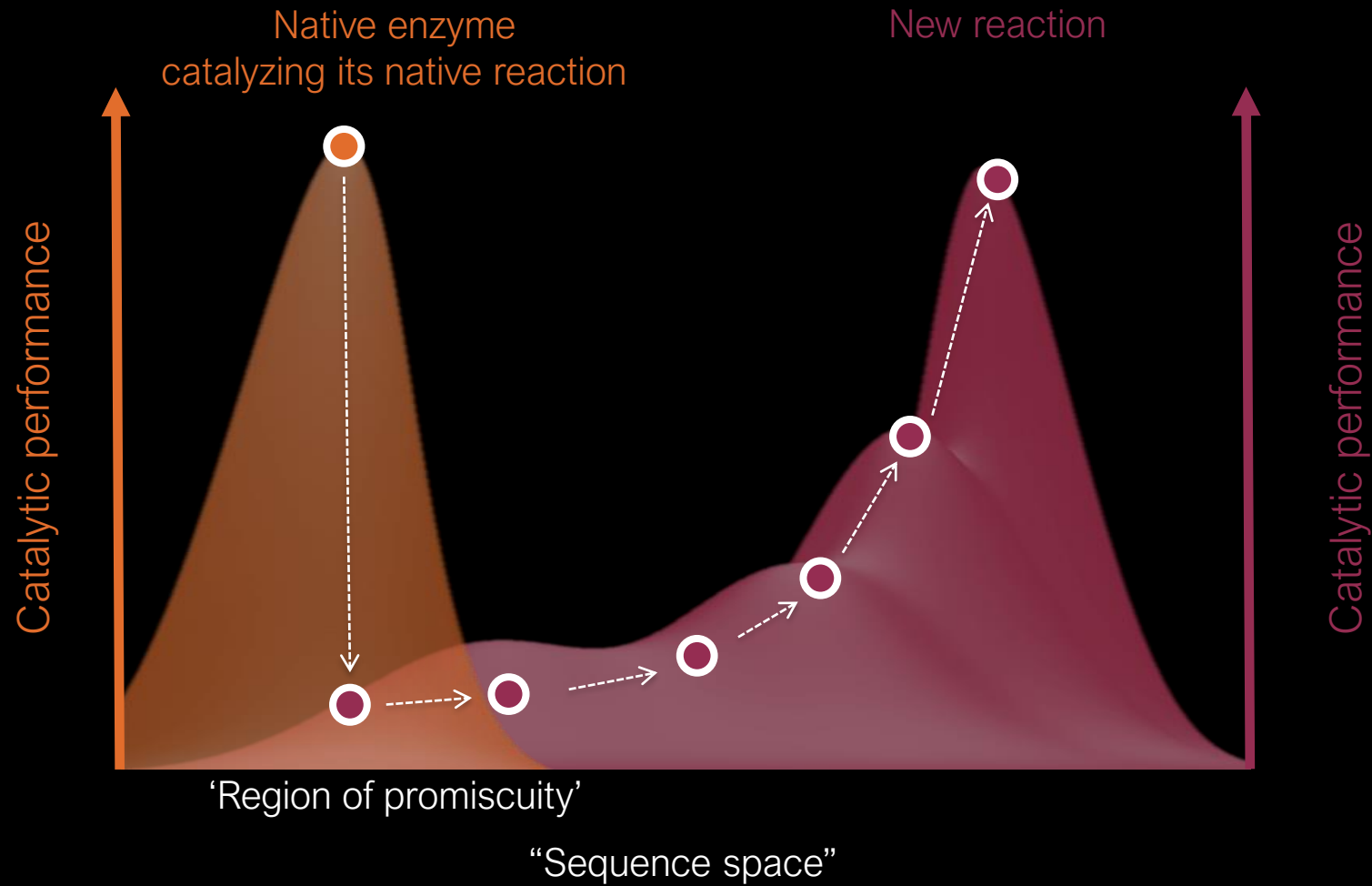
Both:

- ✗ Expensive starting materials

Biocatalysis

- ✓ Selective
- ✓ Easily produced
- ✗ Reversible
- ✗ Poor substrate scope

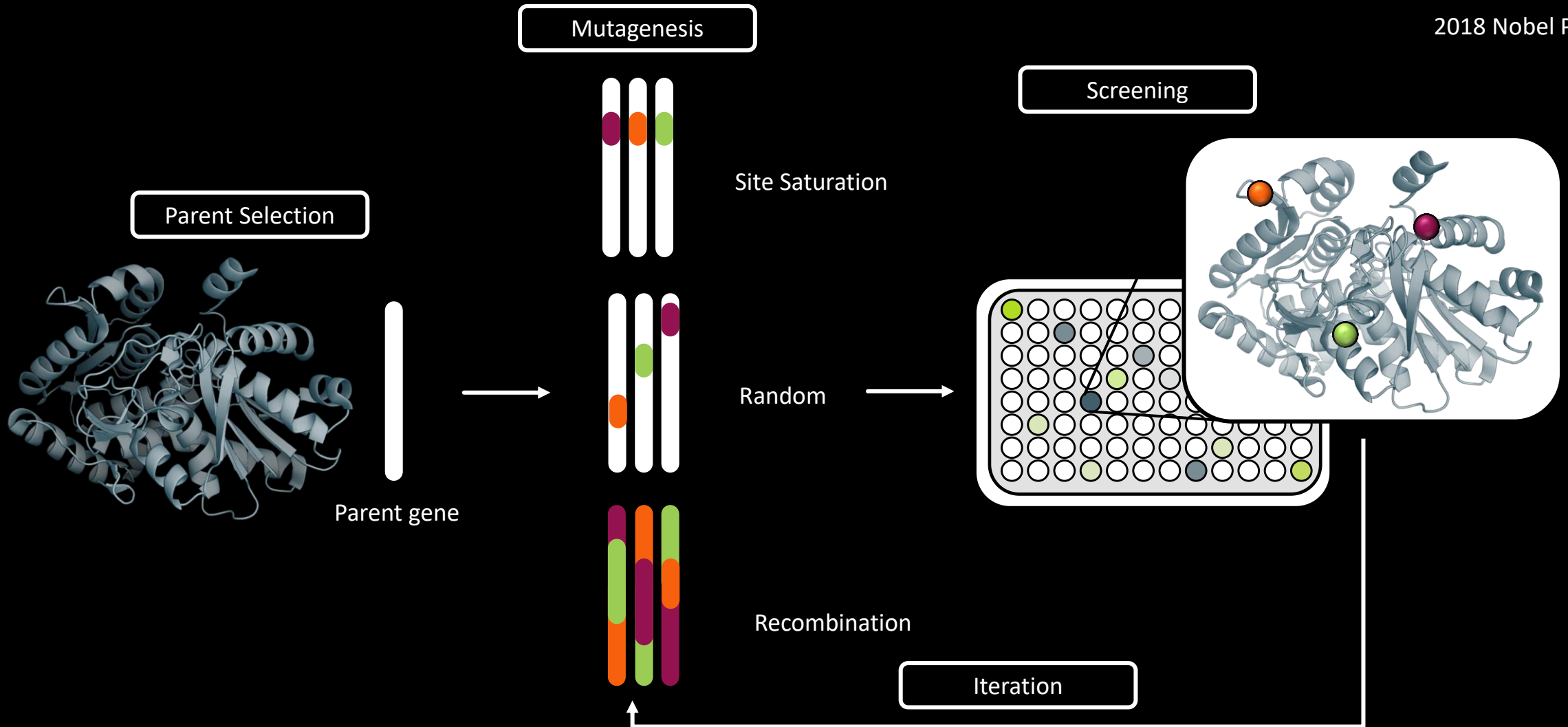
How do new functions arise?



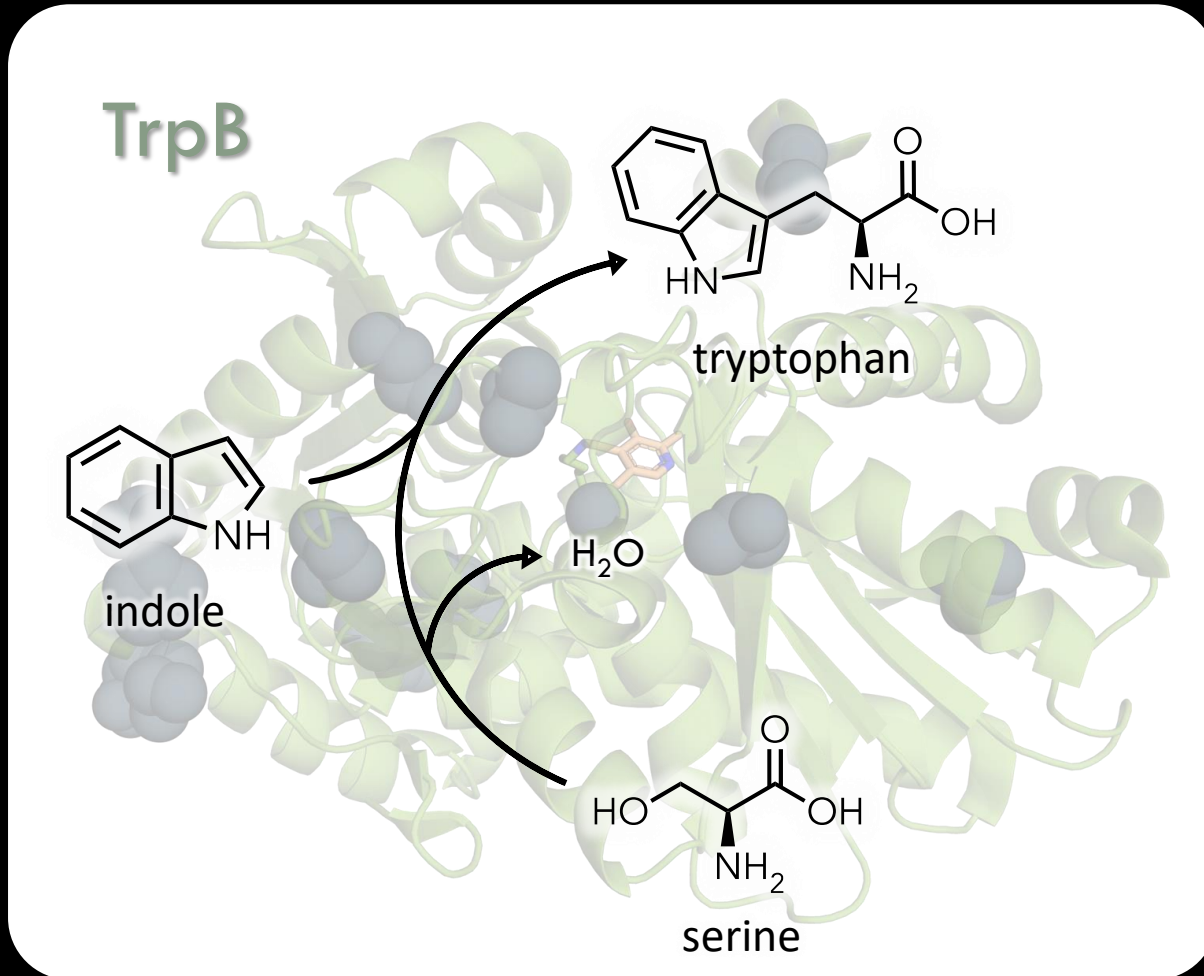
Directed evolution



2018 Nobel Prize



Tryptophan synthase β -subunit



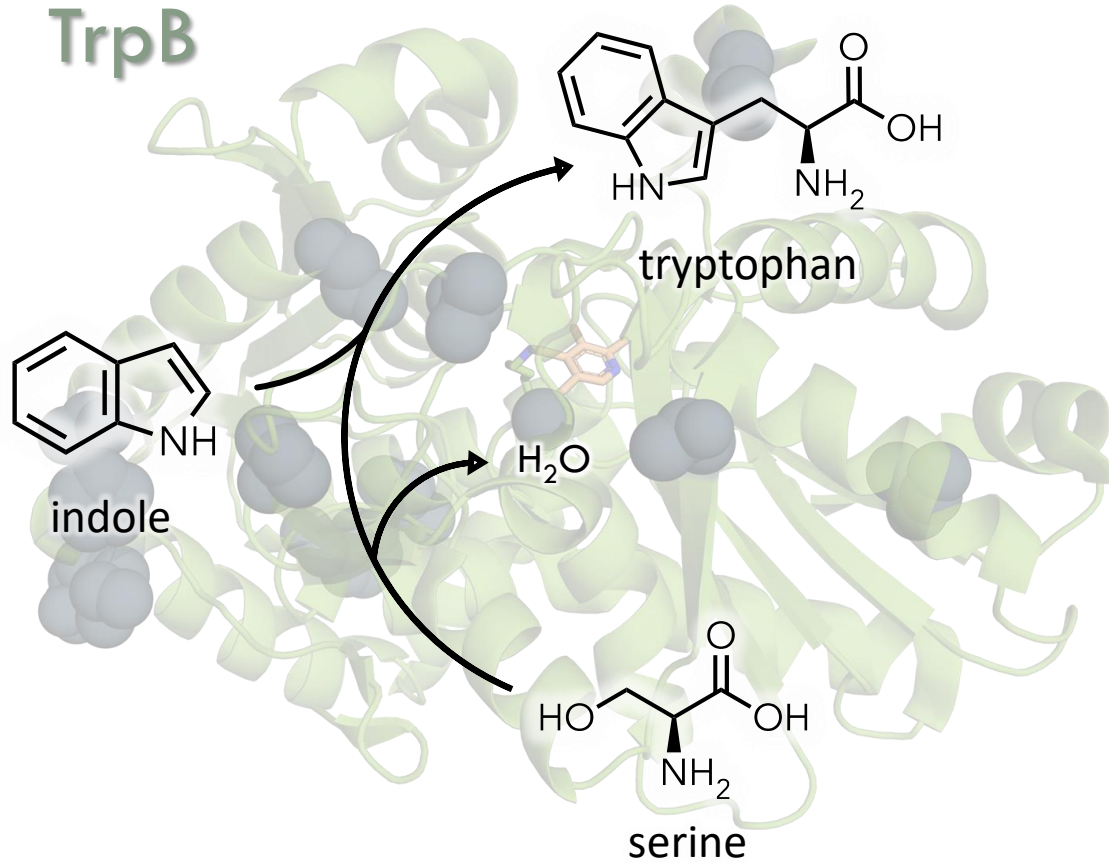
Pyrococcus furiosus



&

Thermotoga maritima
From hydrothermal vents in Italy

Tryptophan synthase β -subunit



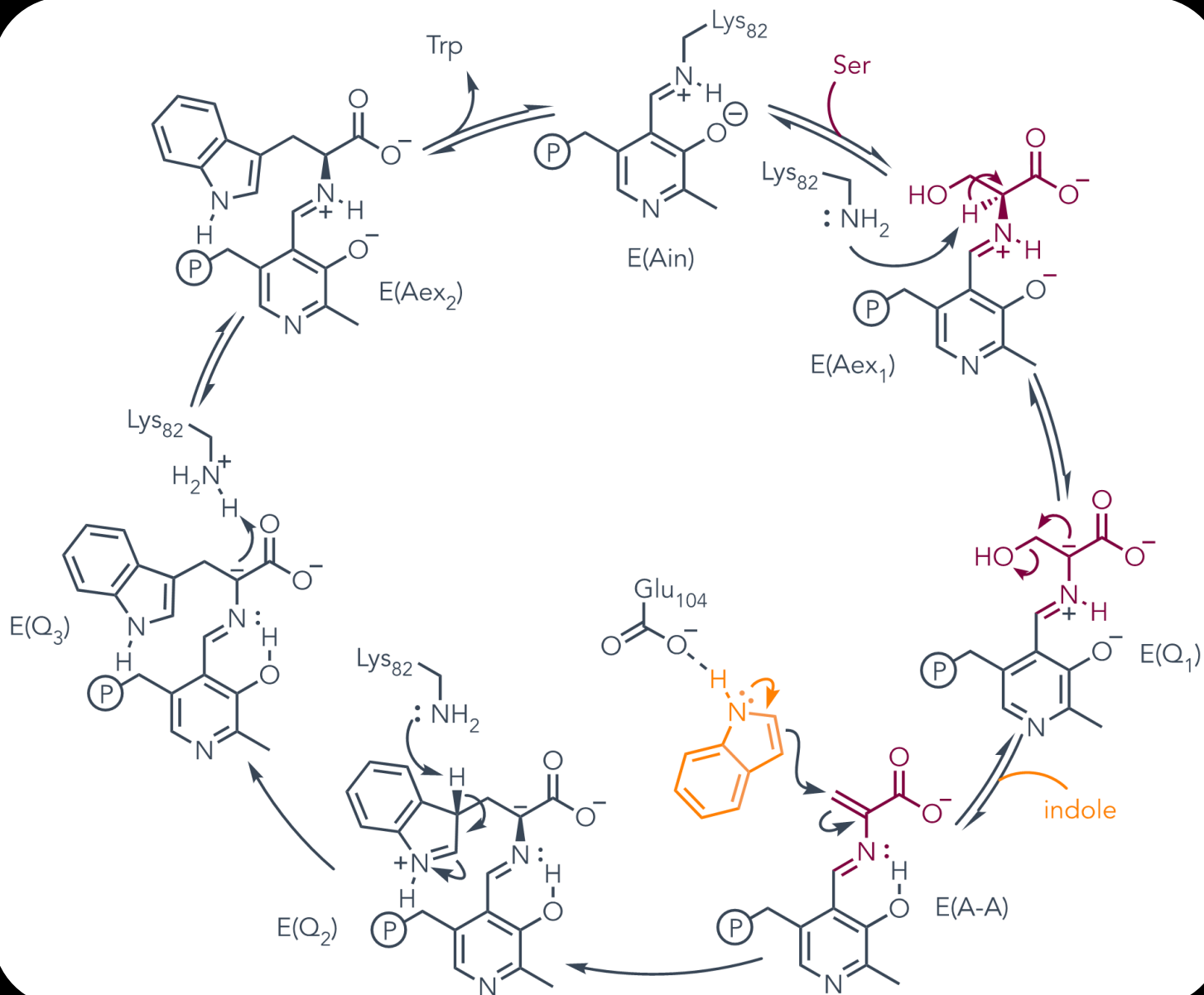
The best of both worlds:

- ✓ Single step
- ✓ Irreversible
- ✓ Simple starting materials
- ✓ Catalytic
- ✓ Aqueous

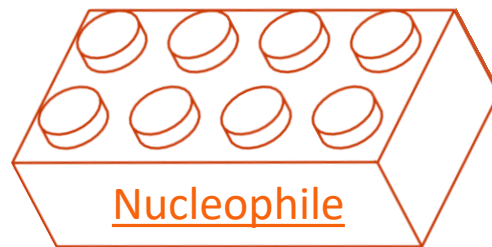
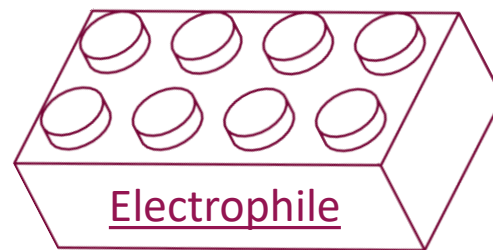
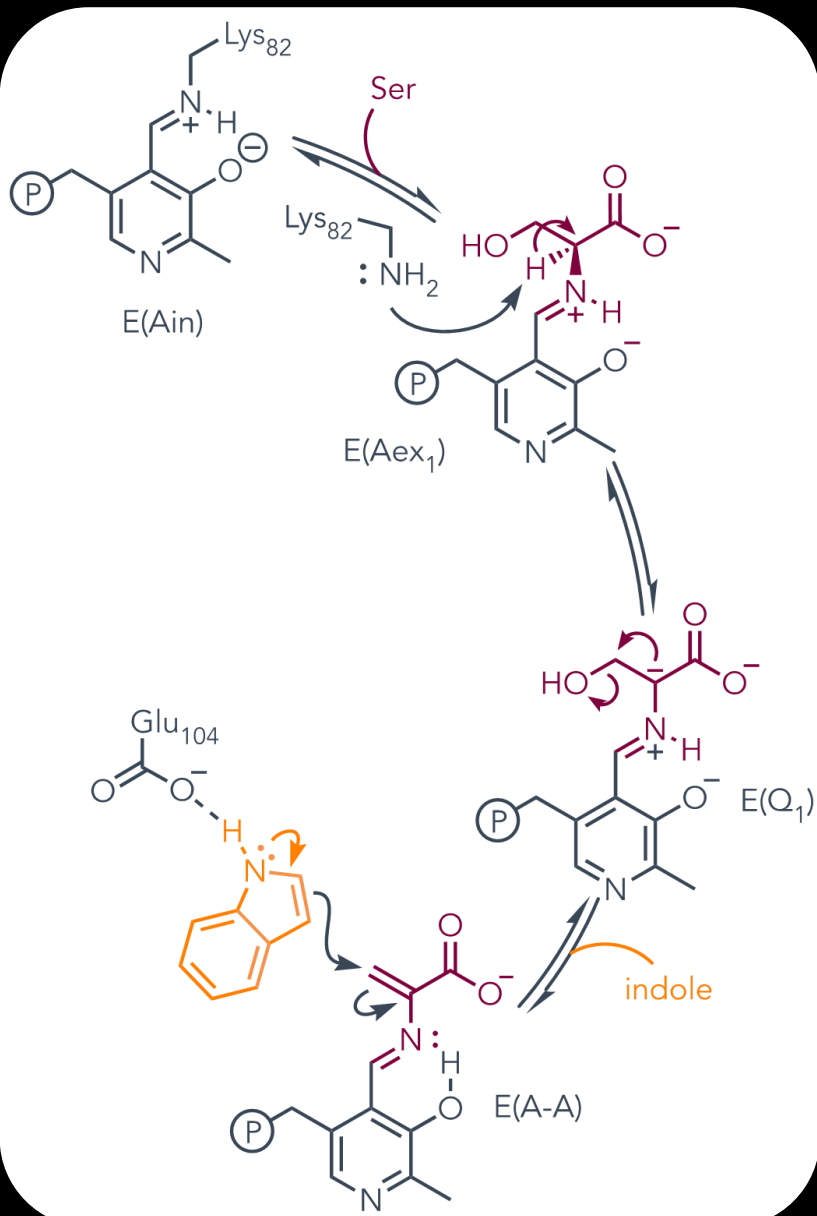
Bonus!

- ✓ ✓ Thermostable
- ✓ ✓ Highly expressing

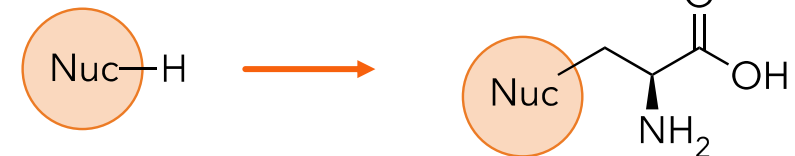
The TrpB catalytic cycle



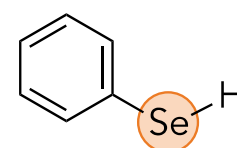
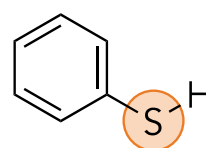
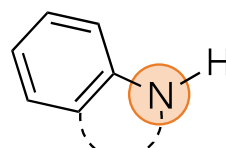
The TrpB reaction is modular



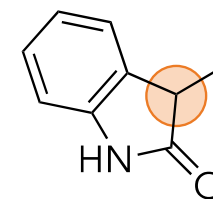
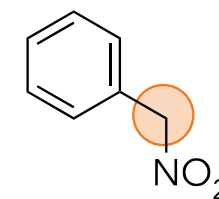
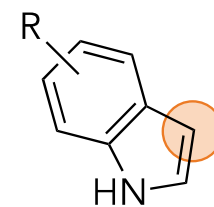
serine +



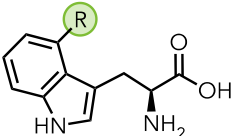
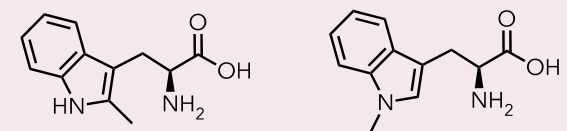
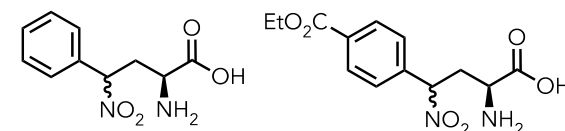
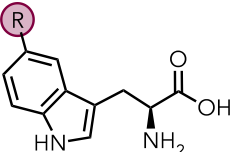
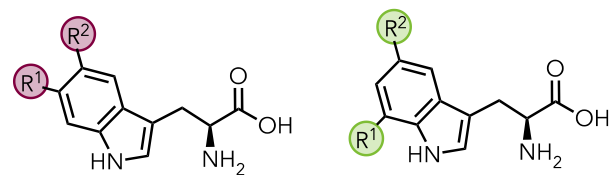
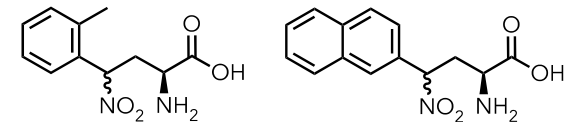
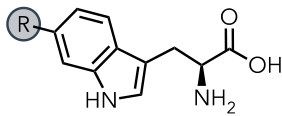


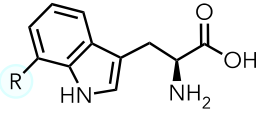
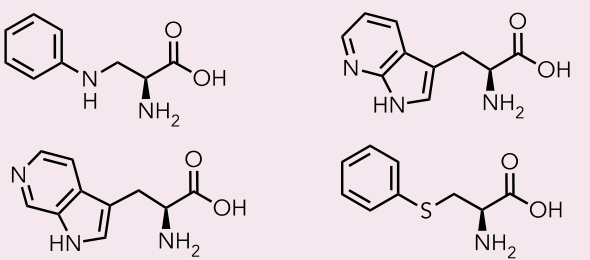
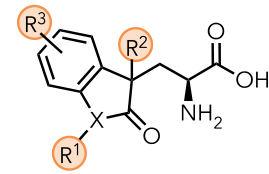
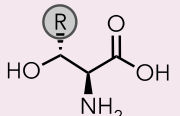
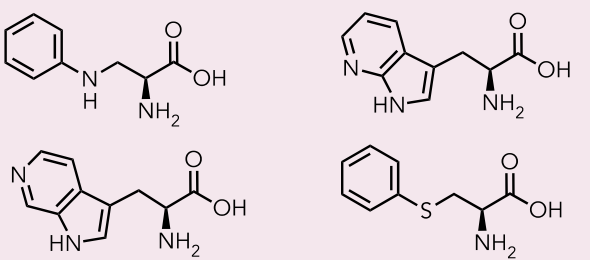
Non-carbon nucleophiles



Carbon nucleophiles



An impressive scope

Substitution	EDG	Halogenated	EWG	1- and 2- substituted	Non-indole carbon nucleophiles
4		CH ₃ OCH ₃ OH	F Cl Br NO ₂ CO ₂ H CN CONH ₂		
5		CH ₃ OCH ₃ OH	F Cl Br I NO ₂ CHO CN CONH ₂ B(OH) ₂		
6		CH ₃ OCH ₃ OH	F Cl Br NO ₂ CN B(OH) ₂		
7		CH ₃ OCH ₃ OH	F Cl Br I NO ₂ CN		
 β-substituted electrophiles accepted R = H, methyl, ethyl, propyl				Other classes of α-amino acids: 	

R¹ = CH₃, H ,
 R² = methyl, ethyl, propyl, butyl, OH,
 R³ = Cl, FI

Buller AR, Brinkmann-Chen S, Romney DK, Herger M, Murciano-Calles J, Arnold FH *Proc Natl Acad Sci*. 2015;112(47):14599-14604.

Romney DK, Murciano-Calles J, Wehrmüller JE, Arnold FH. *J Am Chem Soc*. 2017;139(31):10769-10776.

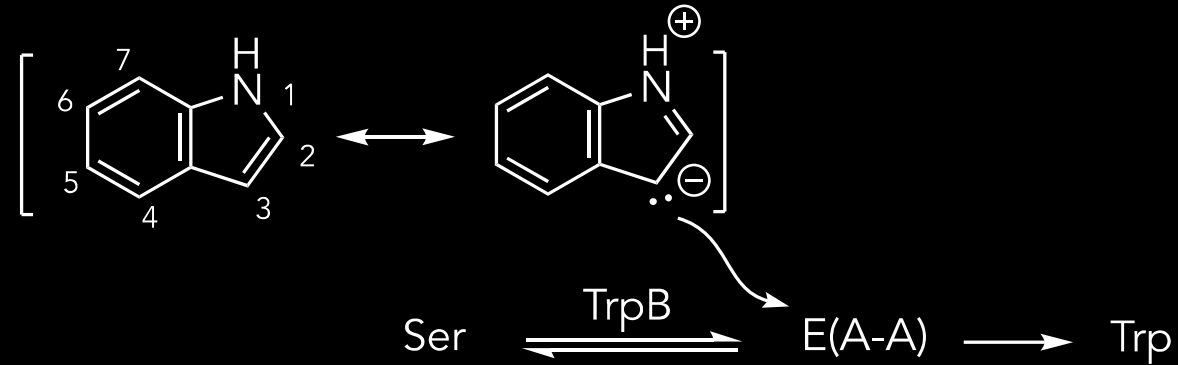
Herger M, van Roye P, Romney DK, Brinkmann-Chen S, Buller AR, Arnold FH. *J Am Chem Soc*. 2016;jacs.6b04836.

Boville CE, Scheele RA, Koch P, Brinkmann-Chen S, Buller AR, Arnold FH. *Angew Chemie Int Ed*. 2018.

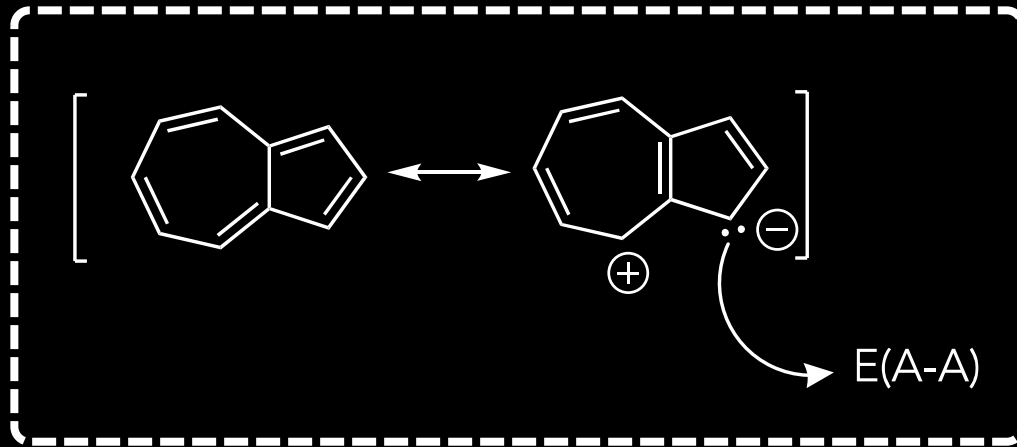
Romney, David K., Nicholas S. Sarai, and Frances H. Arnold. *ACS Catalysis* 2019.

Dick, Markus, et al., *J Am Chem Soc*. 141.50 (2019): 19817-19822.

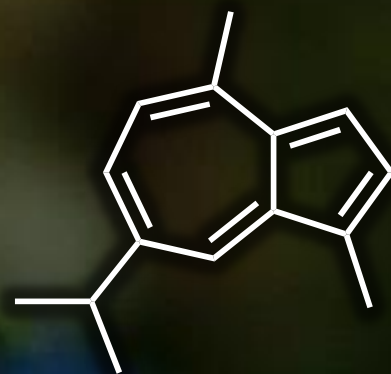
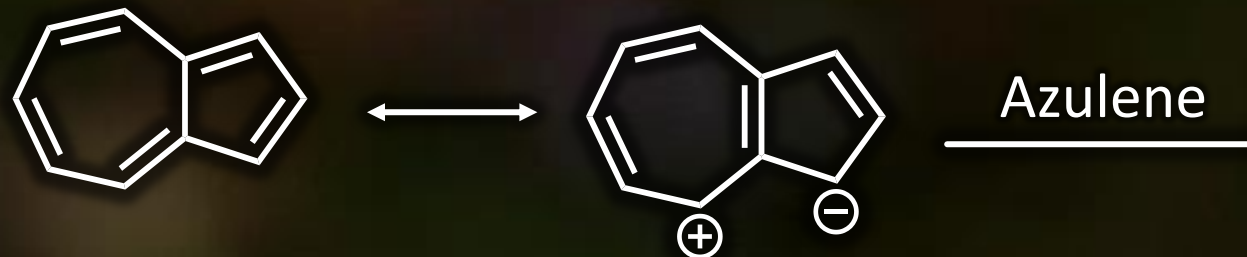
What non-indole carbon nucleophiles could TrpB react with?



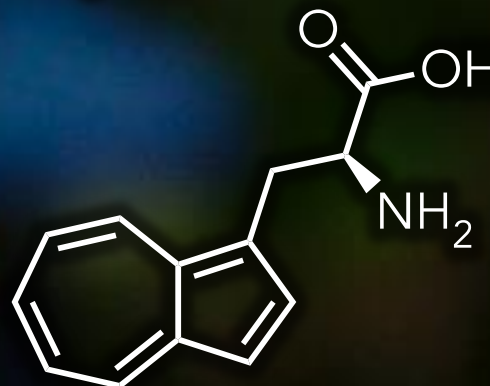
Azulene



The beauty and utility of azulene



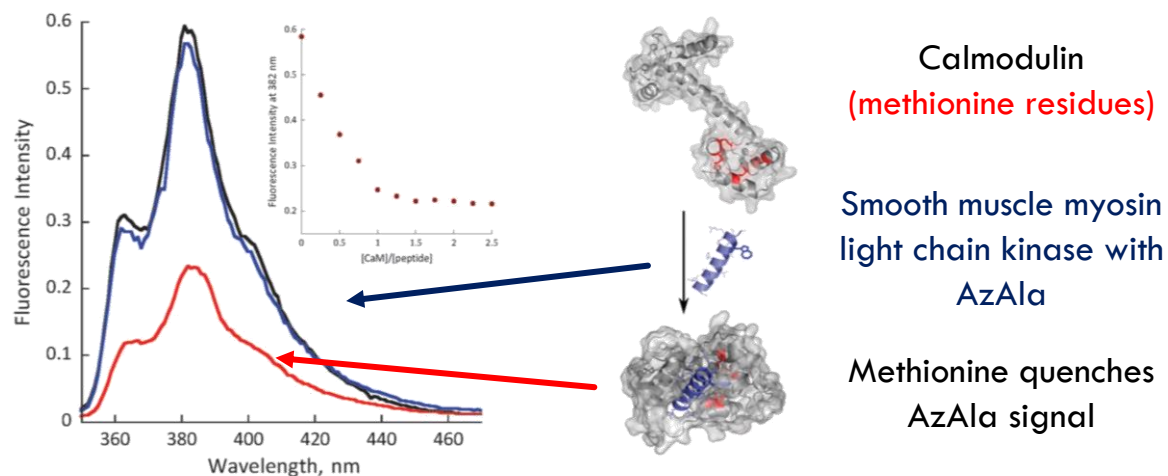
Guaiazulene
Azulene derivative found in mushrooms



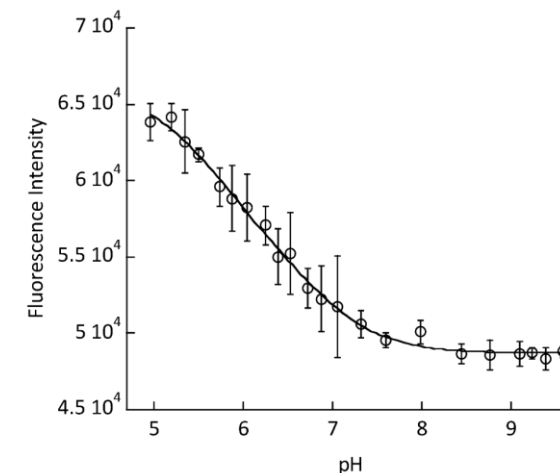
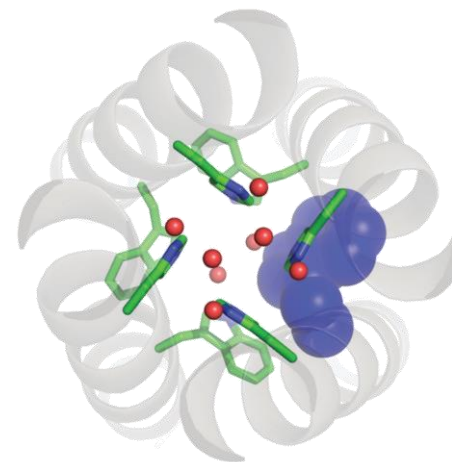
AzAla
Biological fluorescent probe
Tryptophan mimic

Applications of AzAla

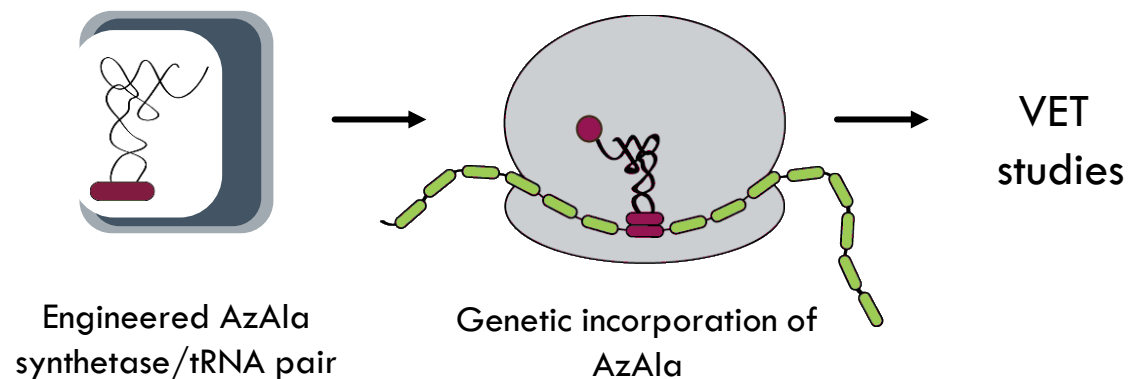
Probe for protein-protein interactions



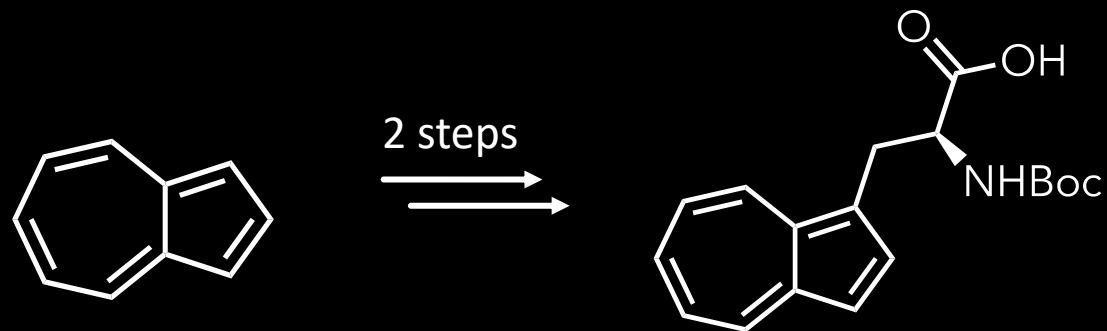
Probe for determination of pK_a of histidine residues



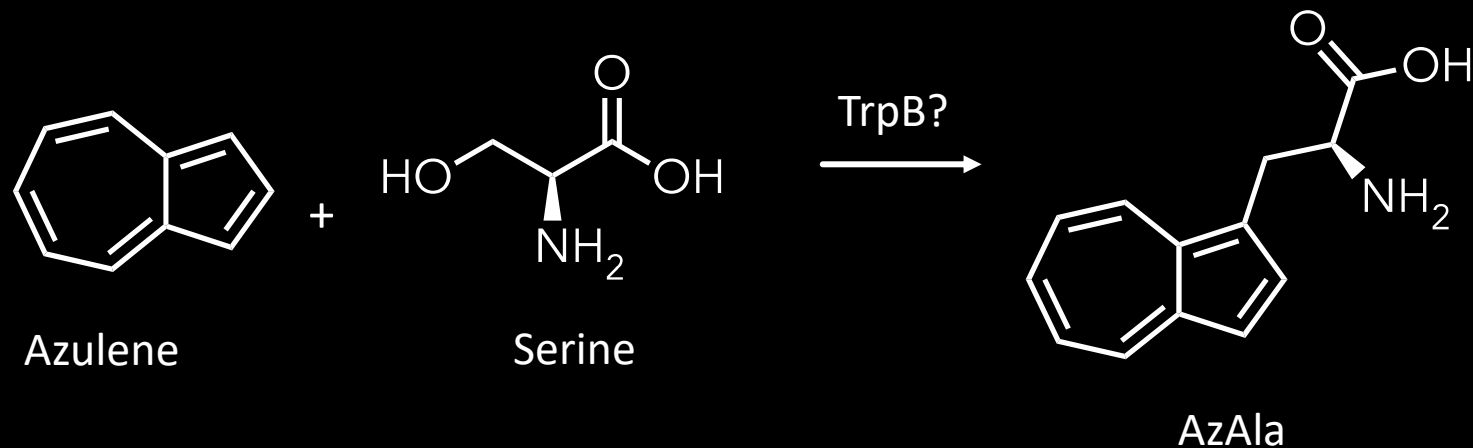
Probe for vibrational energy transfer



Synthesis of AzAla

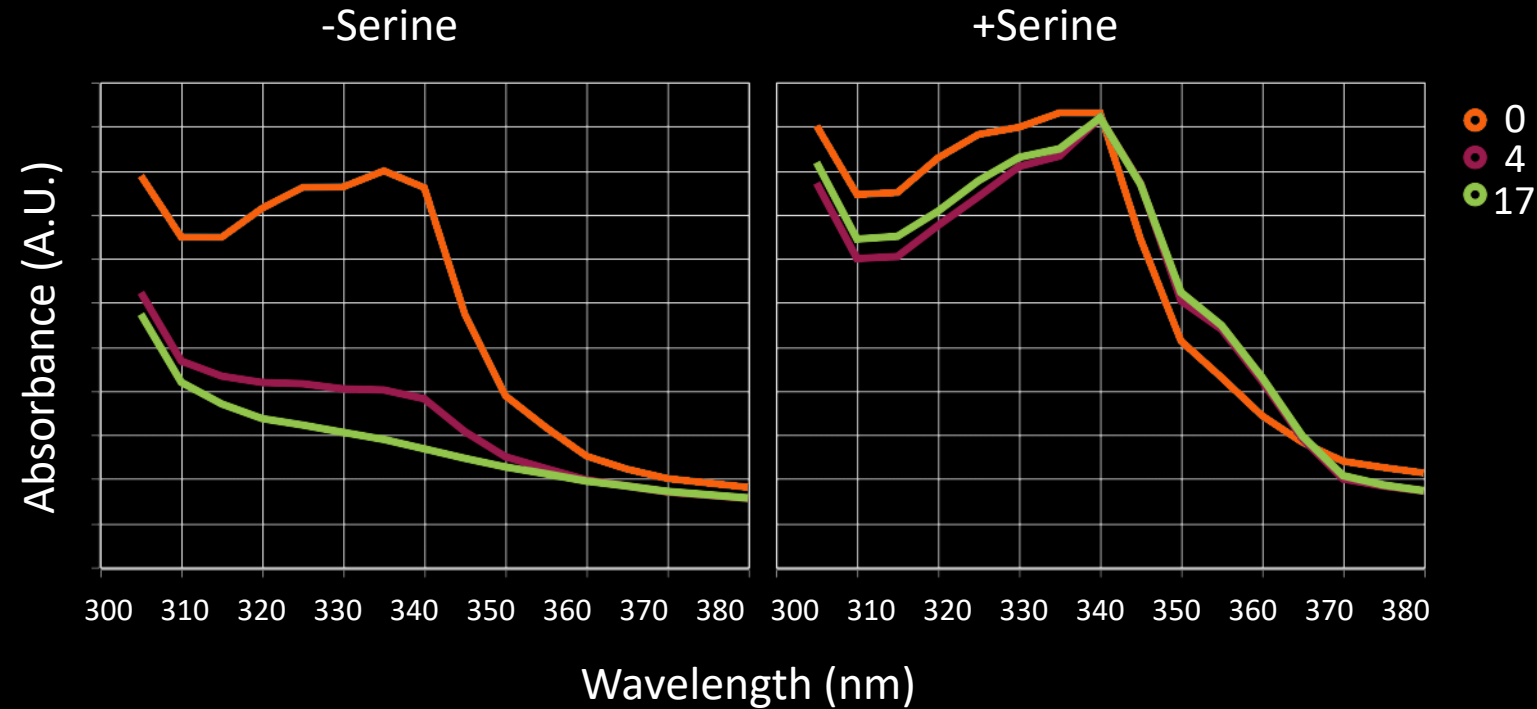
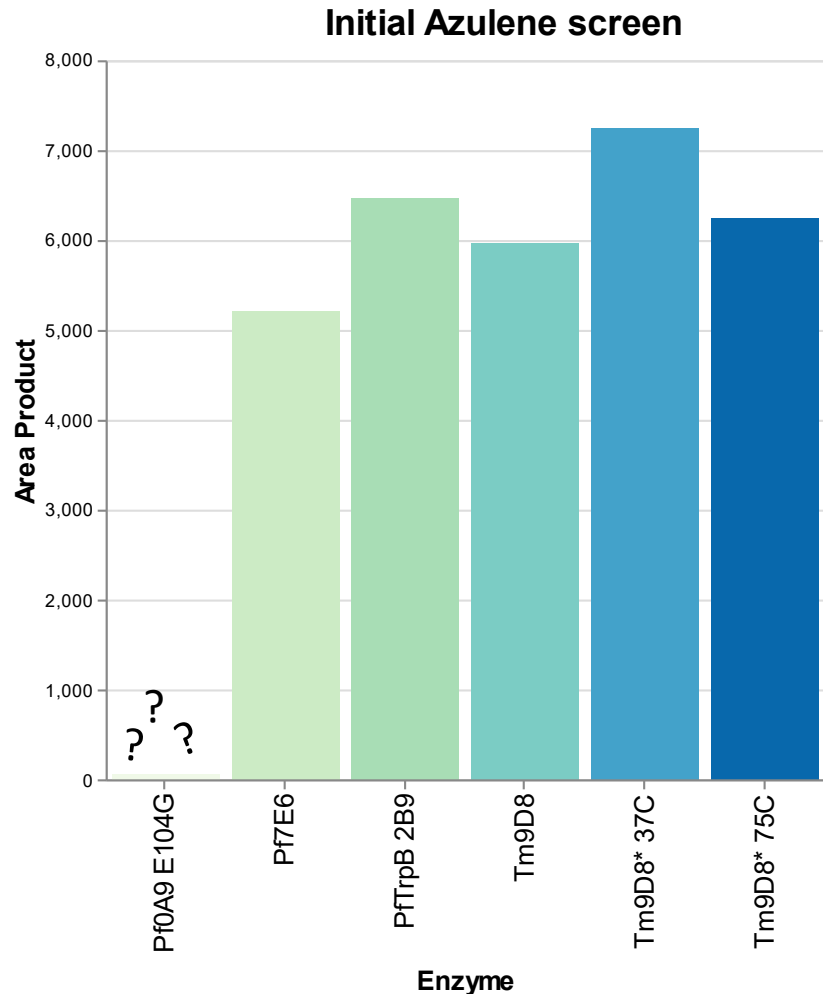


- +2 steps: synthesis of protected amine
- +1 step: deprotection
- Air/water sensitive
- Intermediates unstable at RT

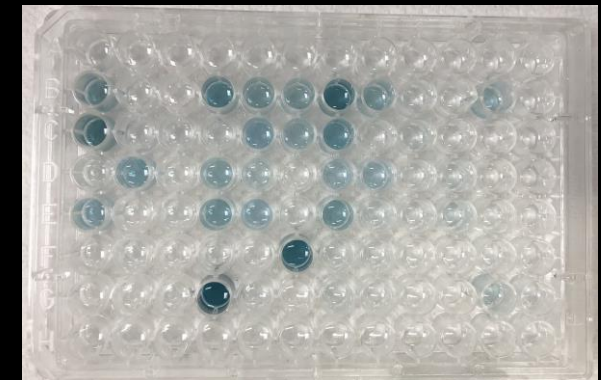


- 1 step
- No protection/deprotection
- Enzymatic
- Not air/water/time sensitive

Finding activity and developing a screen



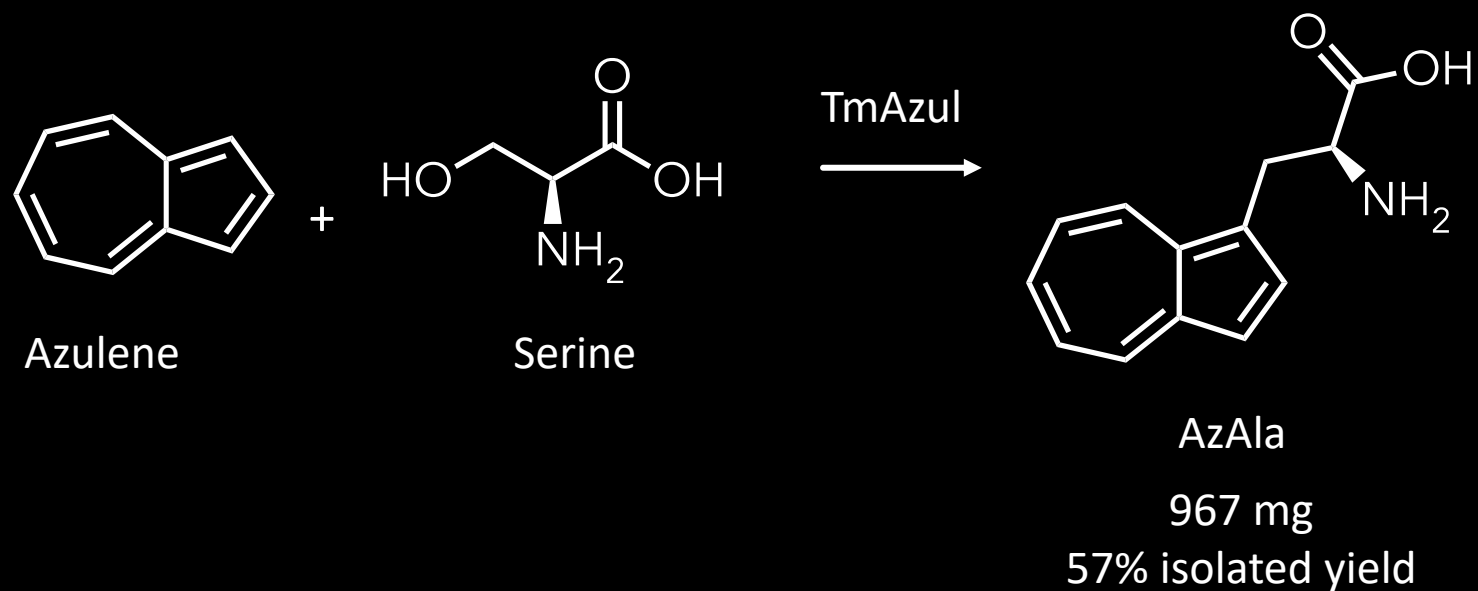
- Azulene evaporates off!
- More vibrant blue wells=better variants
- Only 1 round of evolution was needed
- Tm9D8* + W286R mutation = TmAzul



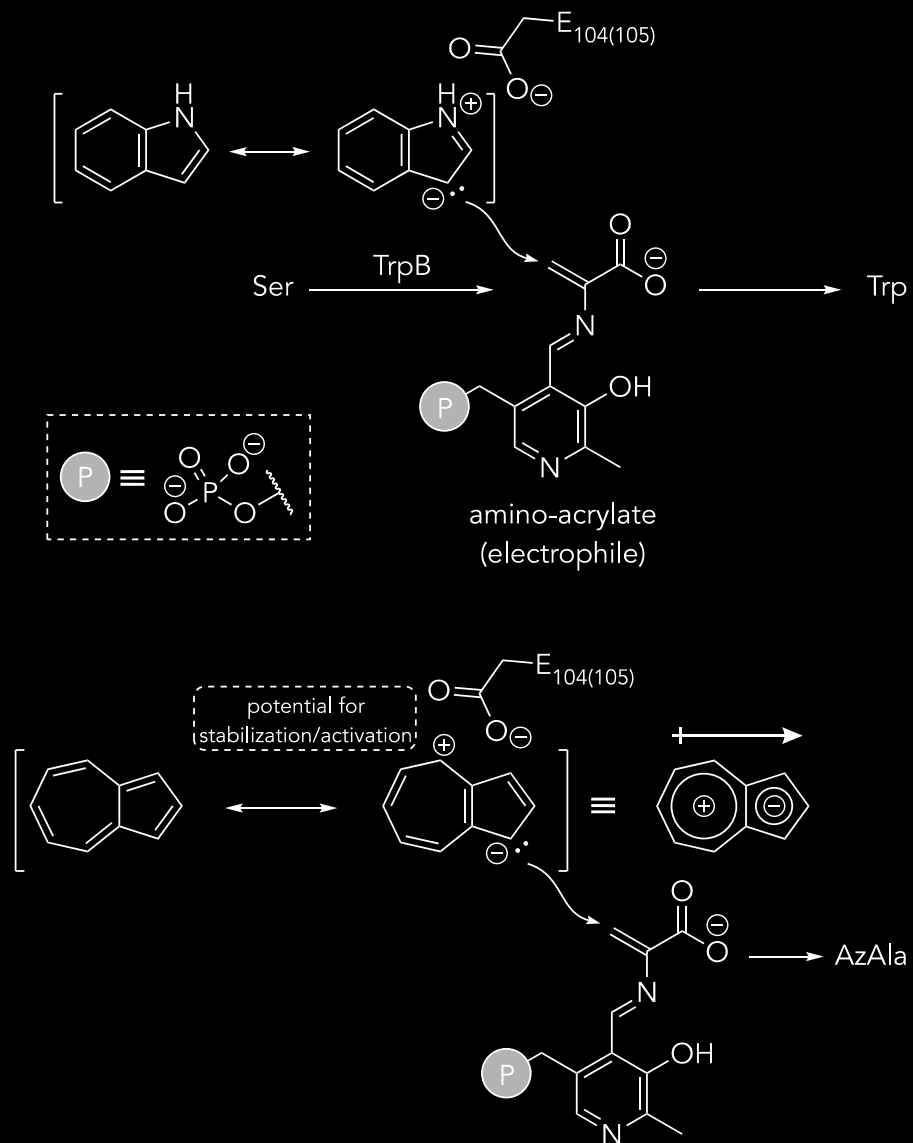
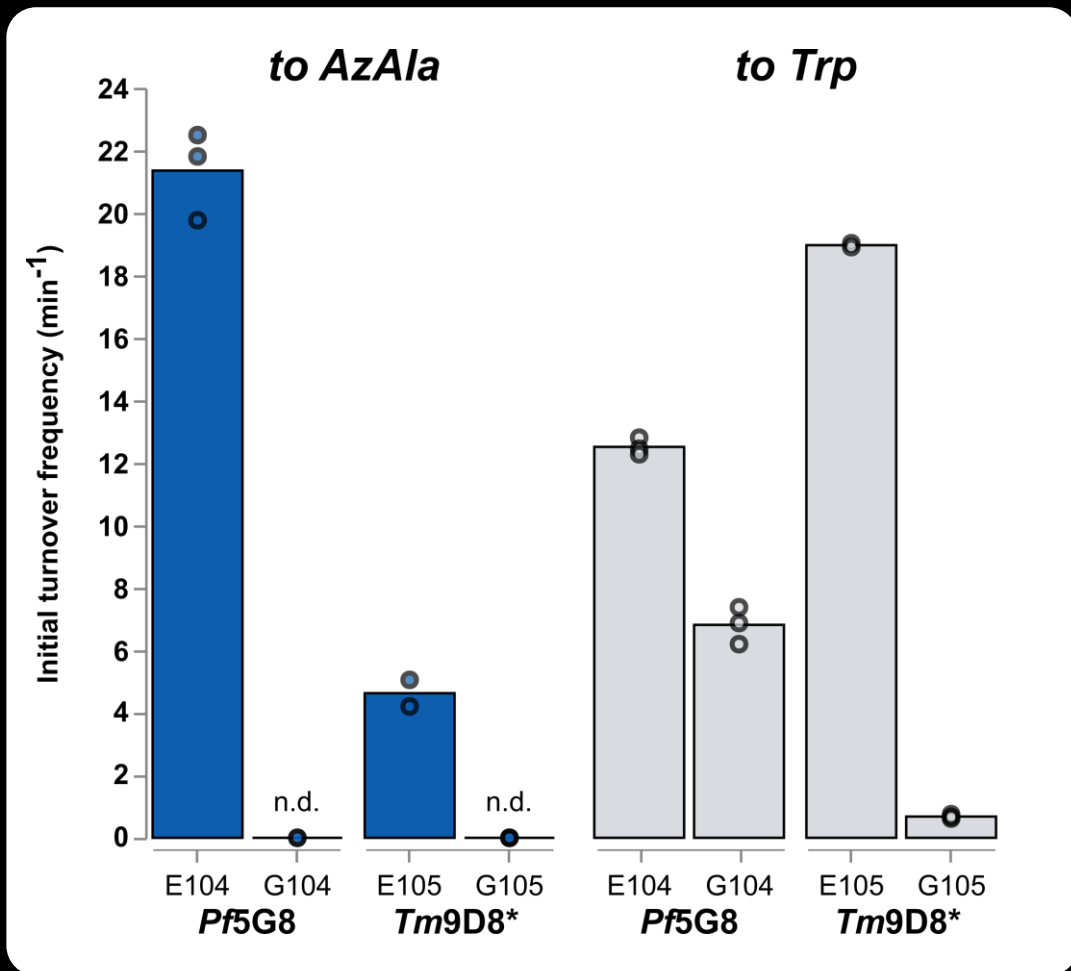


Synthesis of AzAla

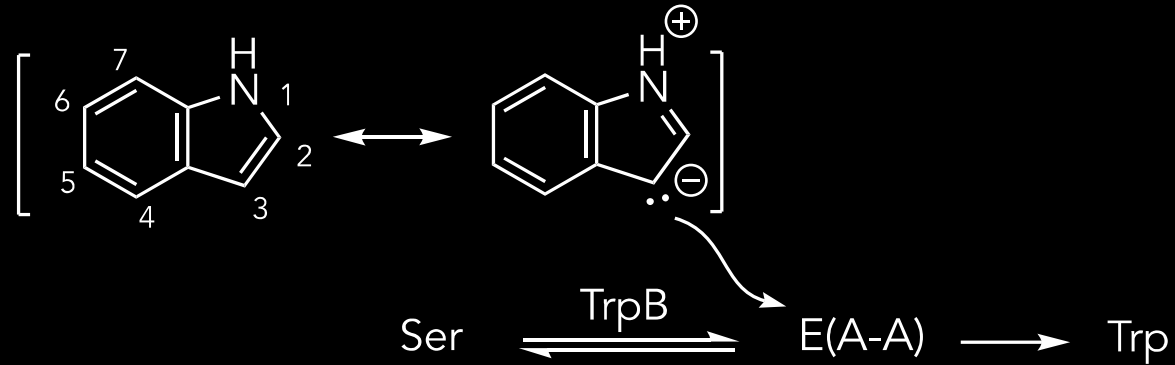
- Optimized scale up conditions
- Optimized purification conditions
 - Fun to do, but lots of work!
 - Easy to track product through every step of purification
- Performed scale up:



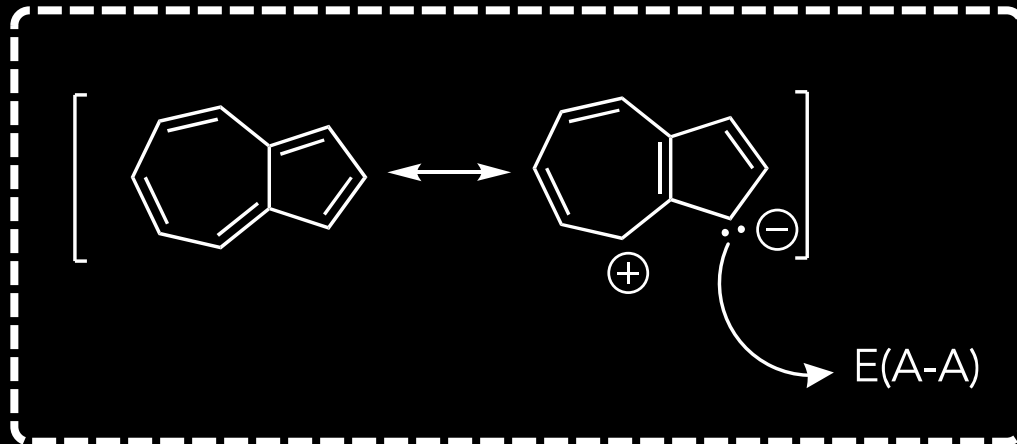
Why does E104(5)G destroy activity?



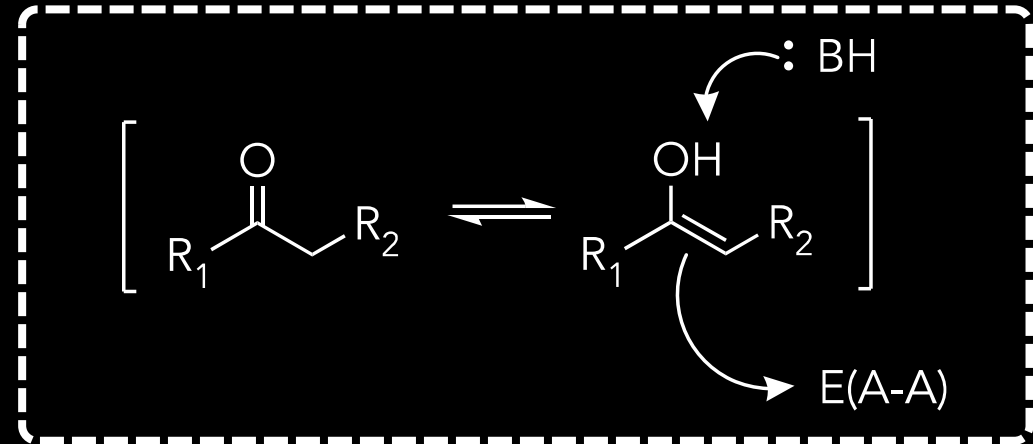
What non-indole carbon nucleophiles could TrpB react with?



Azulene



Ketones



Could TrpB react with *ANY* carbon nucleophile?

Some traditional carbon nucleophiles used in synthetic chemistry are really strong...

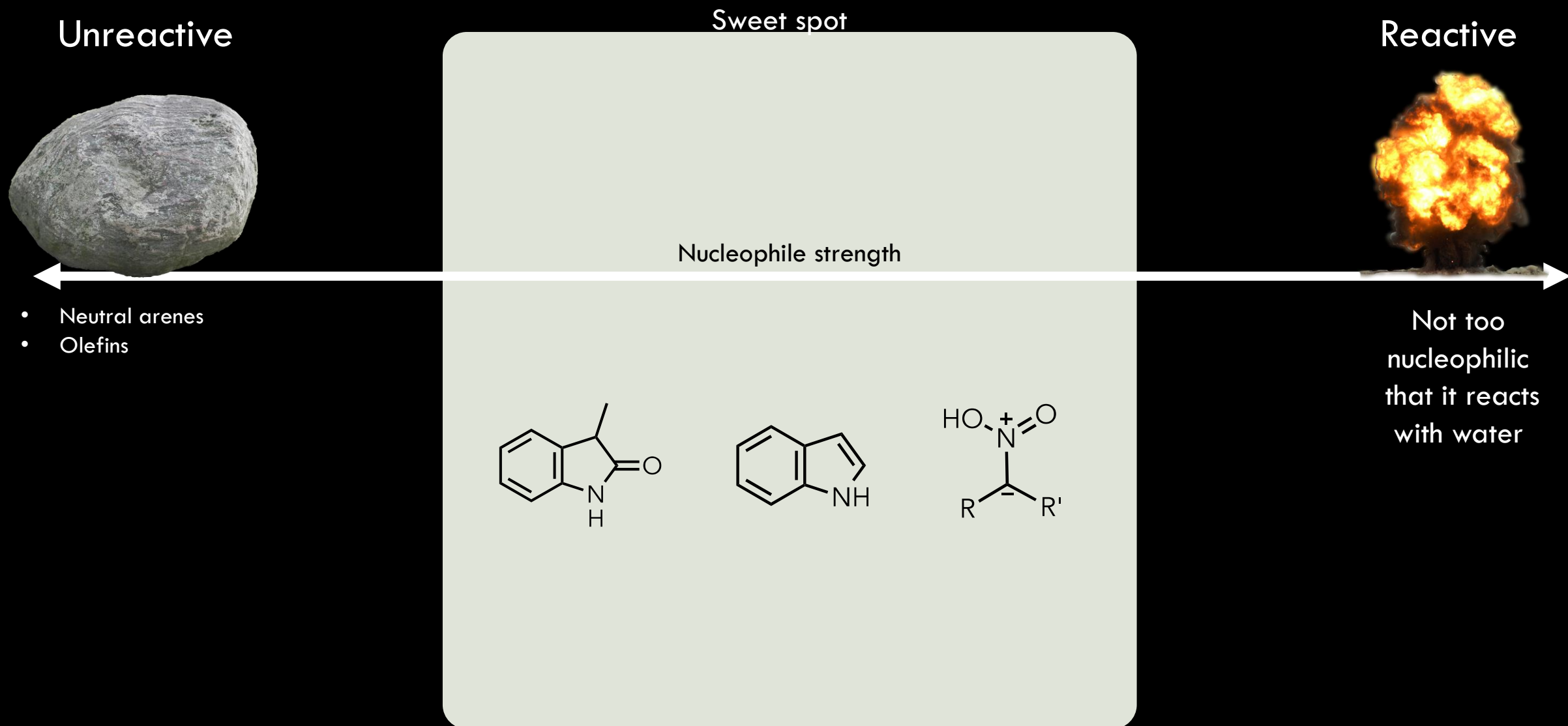


Let's steer clear of N-butyl lithium.



Not *too* nucleophilic that it reacts with water (or air)

What kind of carbon nucleophiles are possible?



Ease of deprotonation

- Ability of carbonyl to stabilize adjacent electron density reflected in pKa values
- $\text{pKa}_{\text{ketones}} \gg \text{pKa}_{\text{water}}$
- Would we be able to reach a **reactive enolate** form in water?

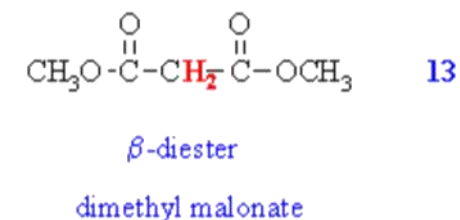
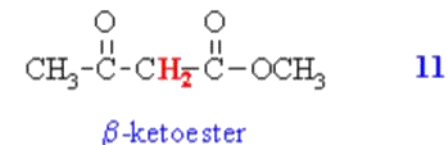
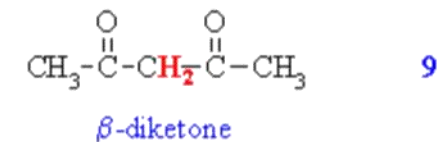
		pKa
carboxylic acid	$\text{HO}-\text{C}(=\text{O})-\text{R}$	5
nitro	CH_3-NO_2	9
amide	$\text{H}_2\text{N}-\text{C}(=\text{O})-\text{R}$	15
aldehyde	$\text{CH}_3-\text{C}(=\text{O})-\text{H}$	17
ketone	$\text{CH}_3-\text{C}(=\text{O})-\text{R}$	20
ester	$\text{CH}_3-\text{C}(=\text{O})-\text{OR}$	25
nitrile	CH_3-CN	25
amide	$\text{CH}_3-\text{C}(=\text{O})-\text{N}(\text{CH}_3)_2$	30
alkane	CH_3-CH_3	~50

ACTIVE METHYLENES

(2 electron withdrawing groups)

H₂O: pKa 14

pKa



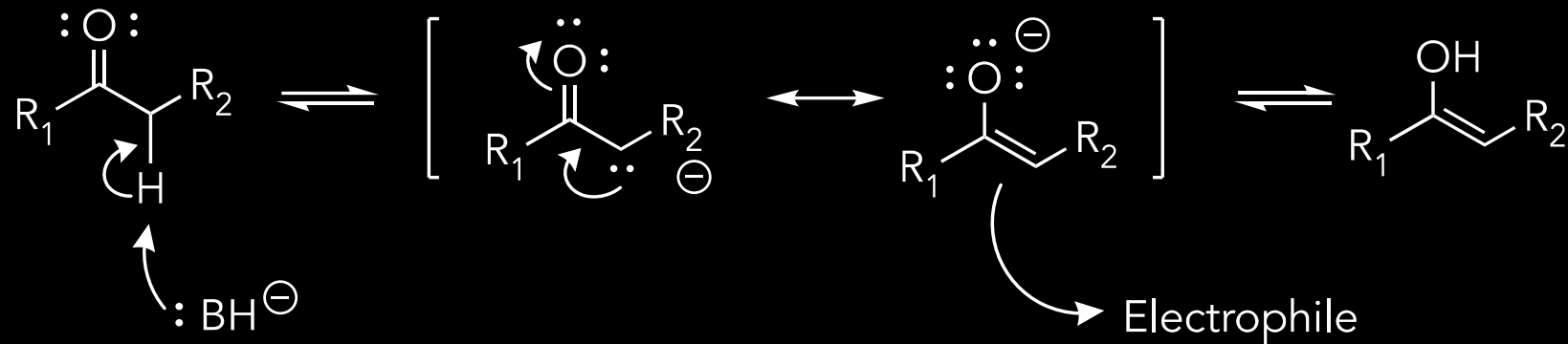
Ketone derived enolates are classic carbon nucleophiles

Ketone

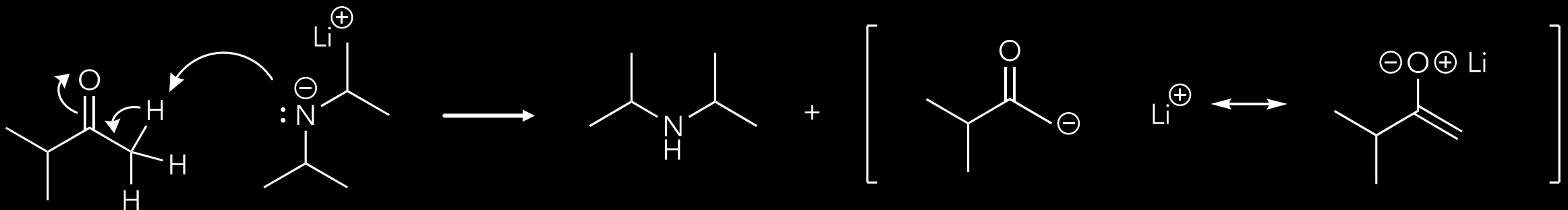
Enolate

(Very fleeting intermediate)

Enol



Example: Lithium diisopropylamide (LDA)



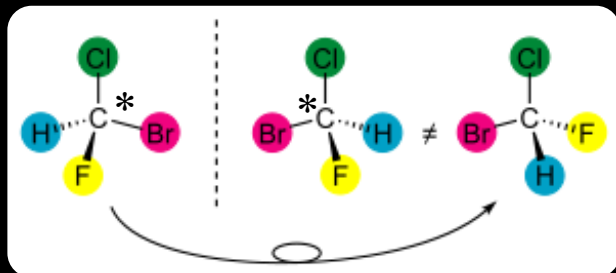
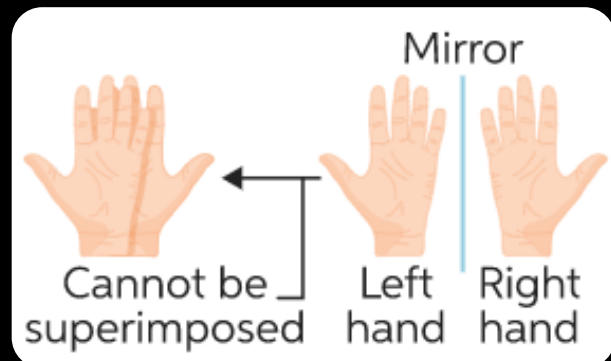
Positively charged lithium stabilizes the negatively charged enolate, making it a persistent species

Easy for chemists to make, difficult to control...

“Direct asymmetric α -C–H alkylation of ketones still unconquered”

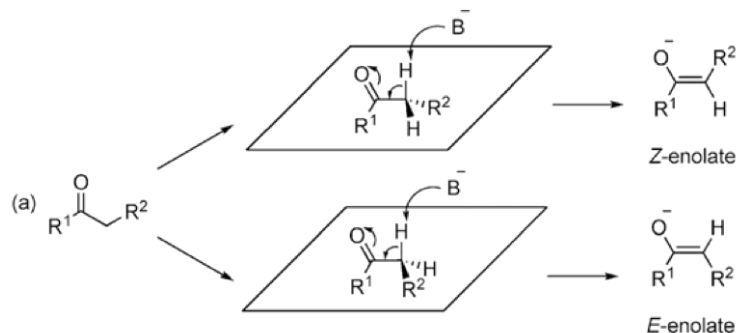
“Asymmetry”

Molecules can be chiral, meaning that they are mirror images that can't be superimposed.

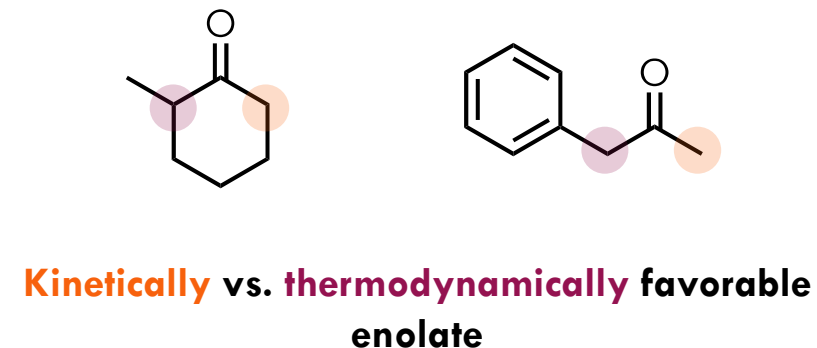


- If a molecule is chiral, it has a “stereocenter” (denoted by a star: *)
- The two mirror images of the molecules are called “stereoisomers”
- Selectivity is crucial: different stereoisomers interact with the body differently

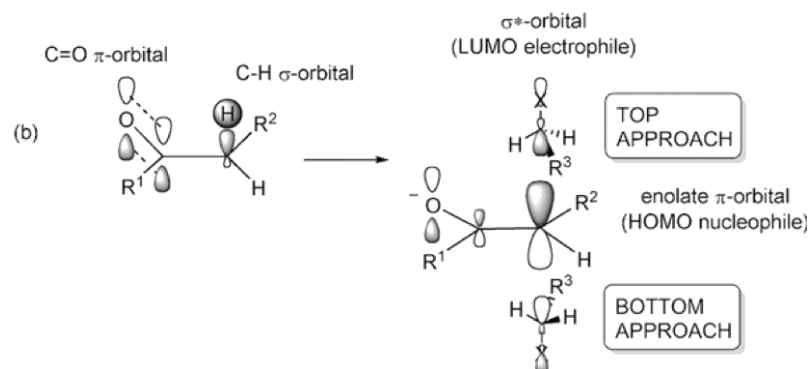
E/Z-enolate selectivity



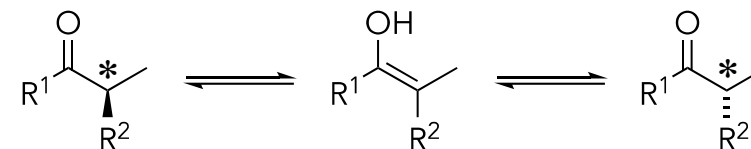
Regioselective enolate formation



Electrophile orbital approach

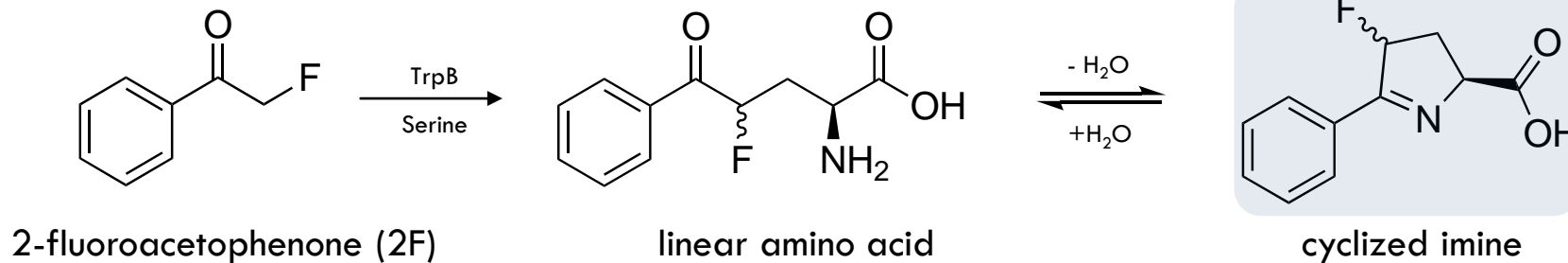
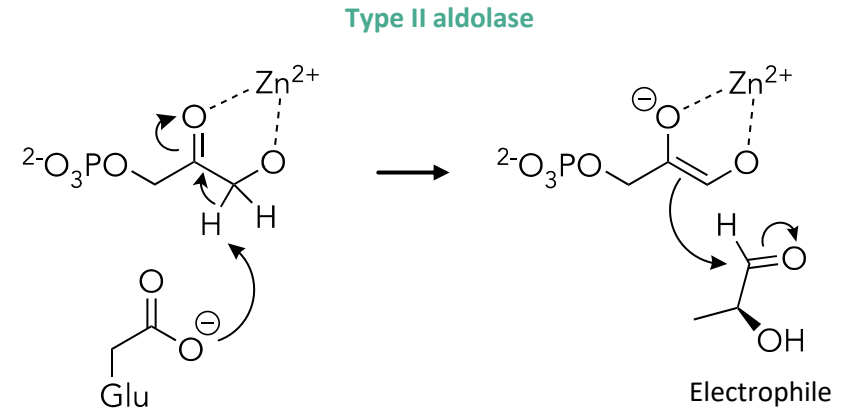
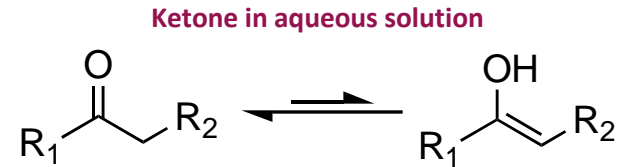


Racemization of new chiral center

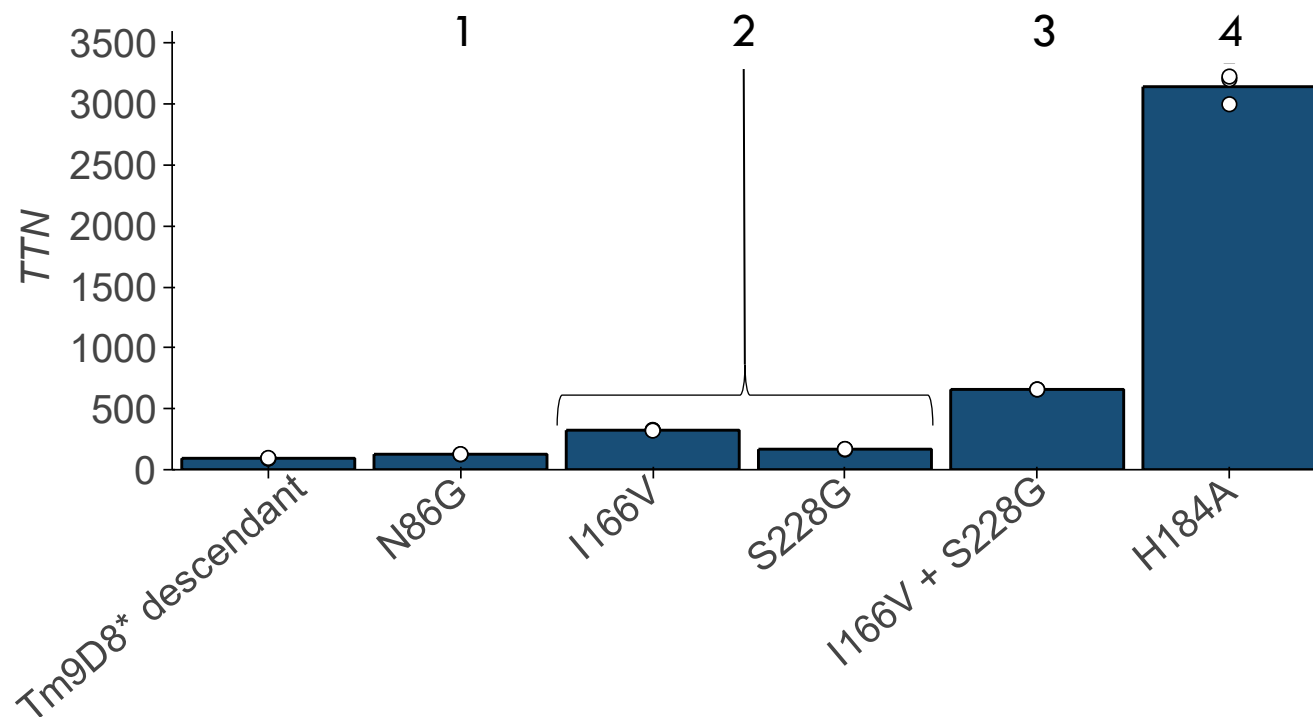
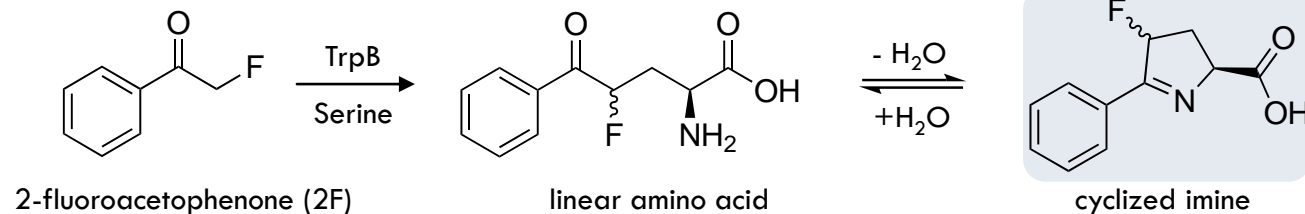


Could TrpB accept ketones as nucleophiles?

- Enols/enolates are **thermodynamically unfavorable** in water
- Very few enzymes that use enolates. Those that do are **very specialized**.



Directed evolution for ketone alkylation



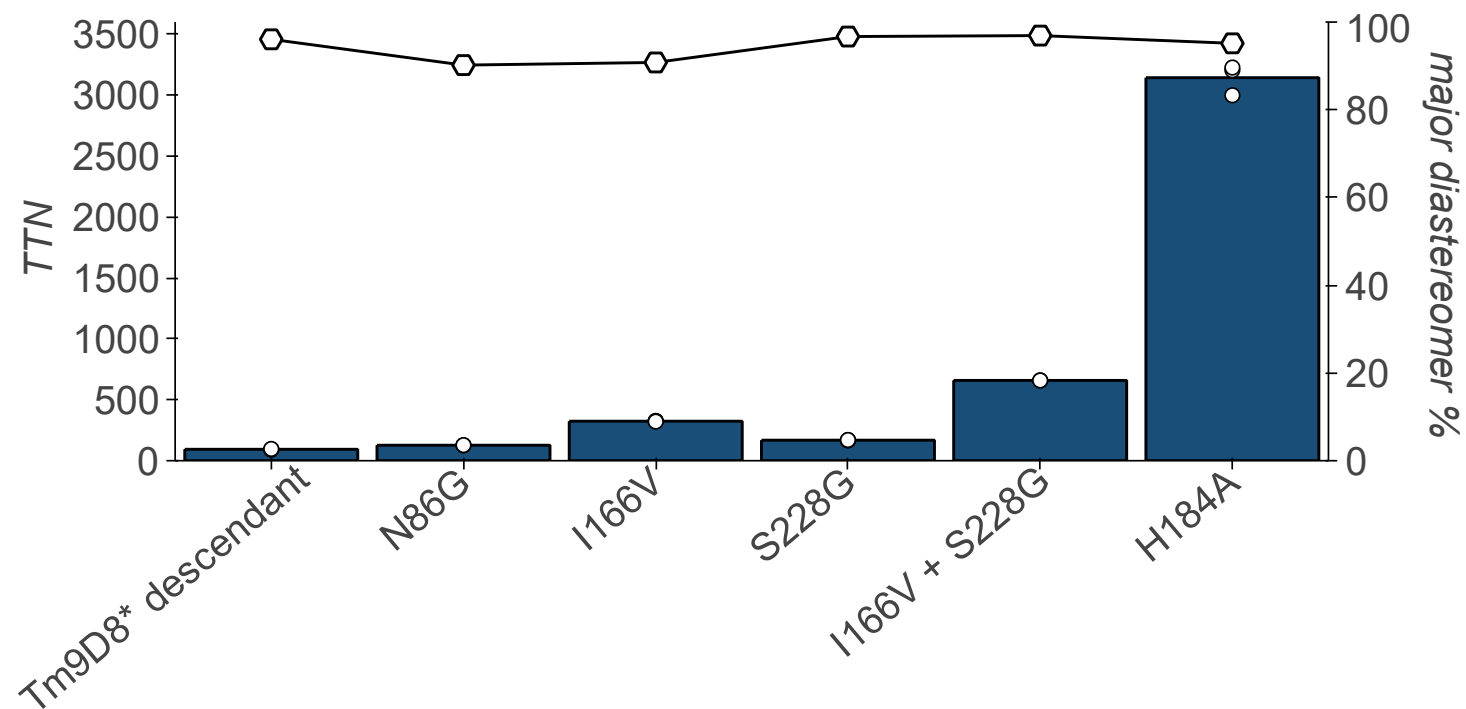
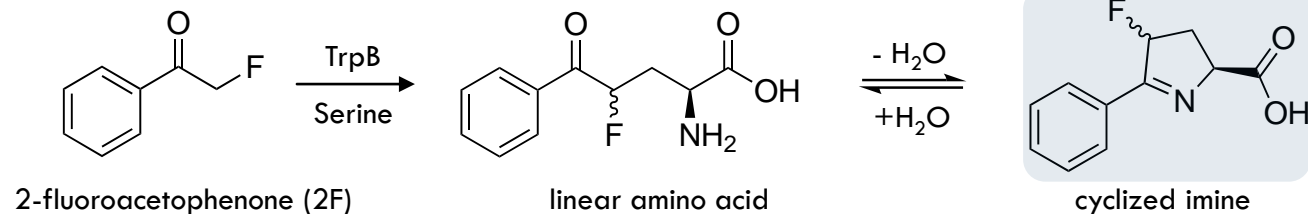
Round 1: Site saturation mutagenesis

Round 2: Random mutagenesis

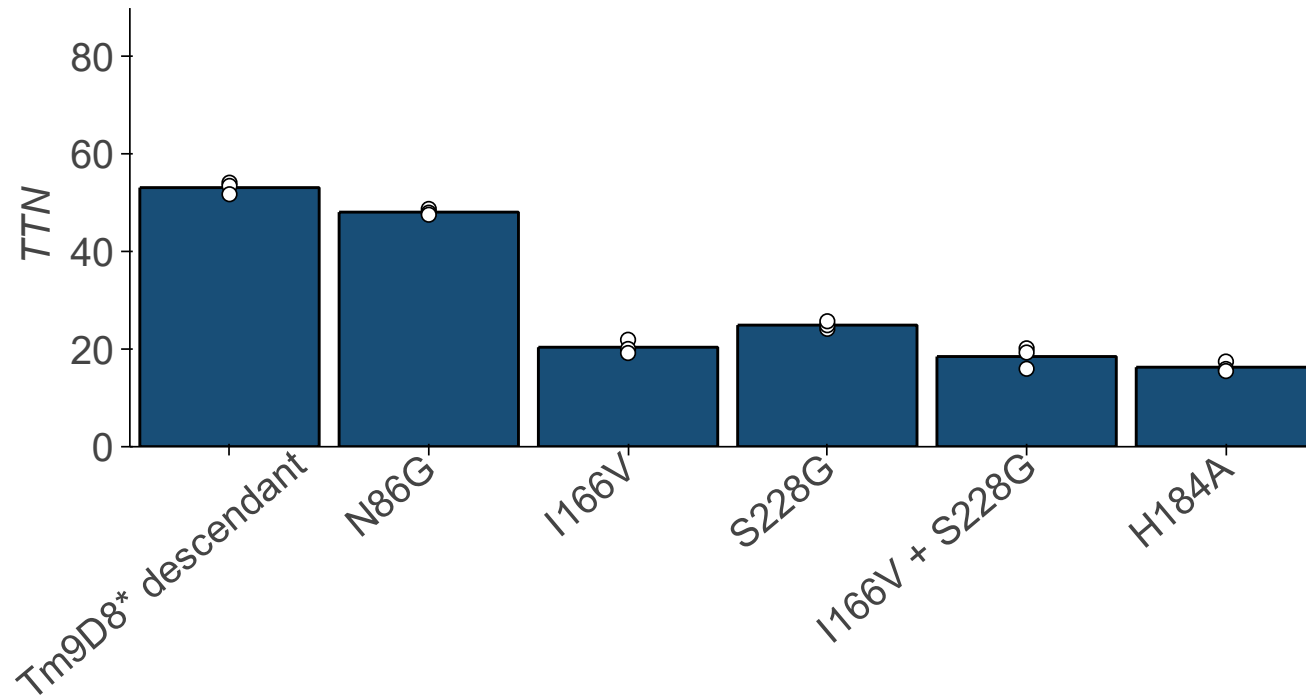
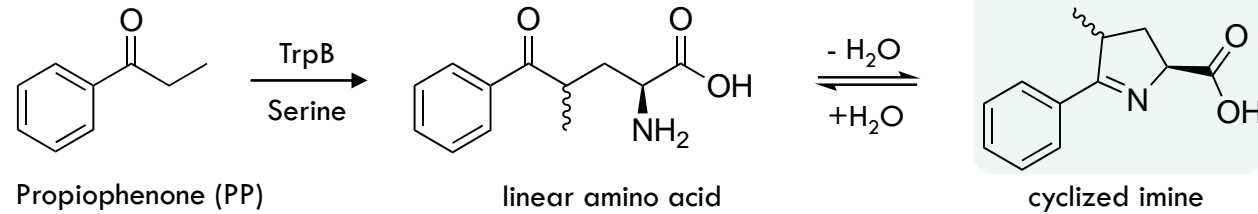
Round 3: Recombination

Round 4: Site directed mutagenesis

TrpB catalyzes asymmetric alkylation!

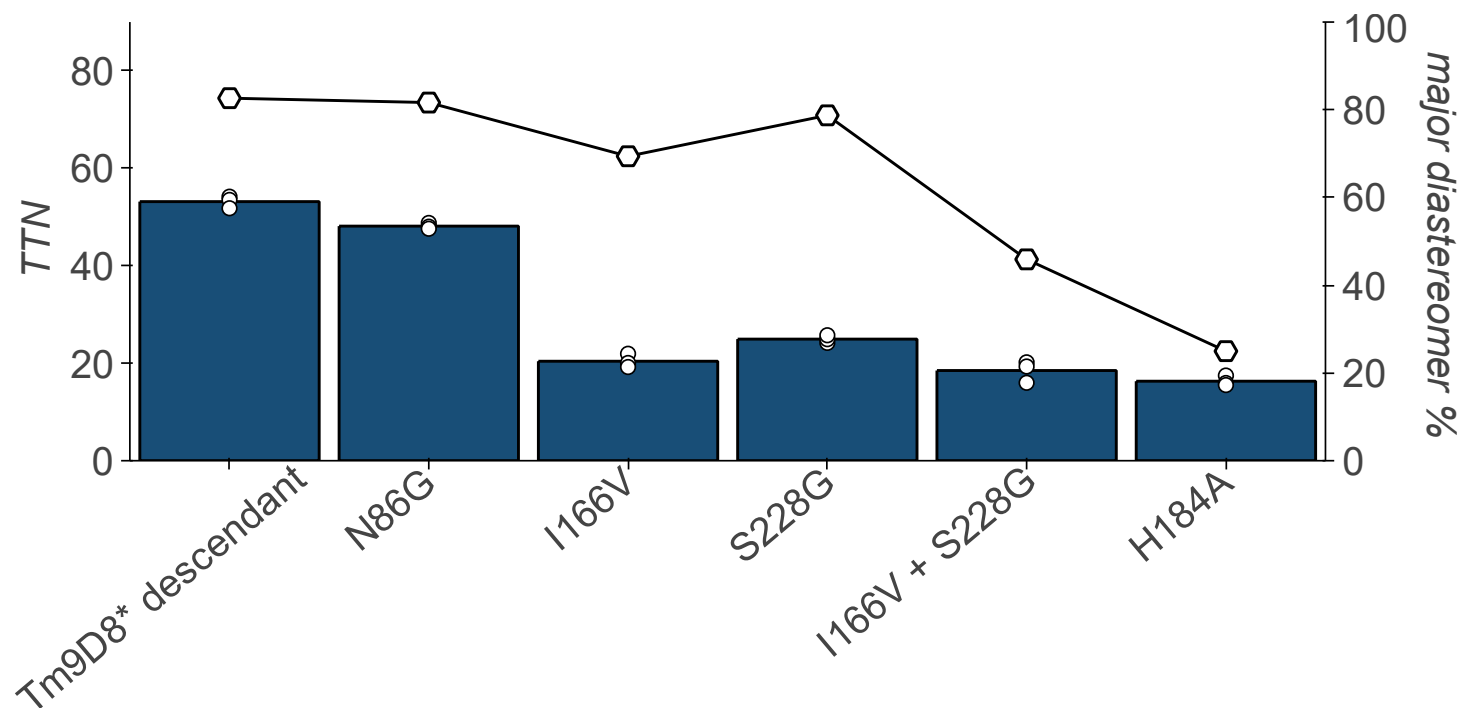
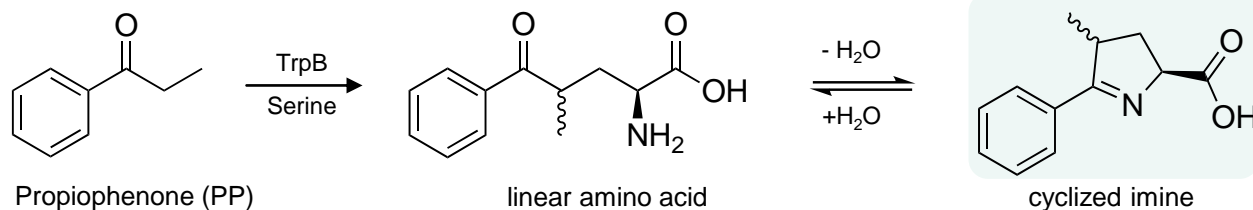


You get what you screen for



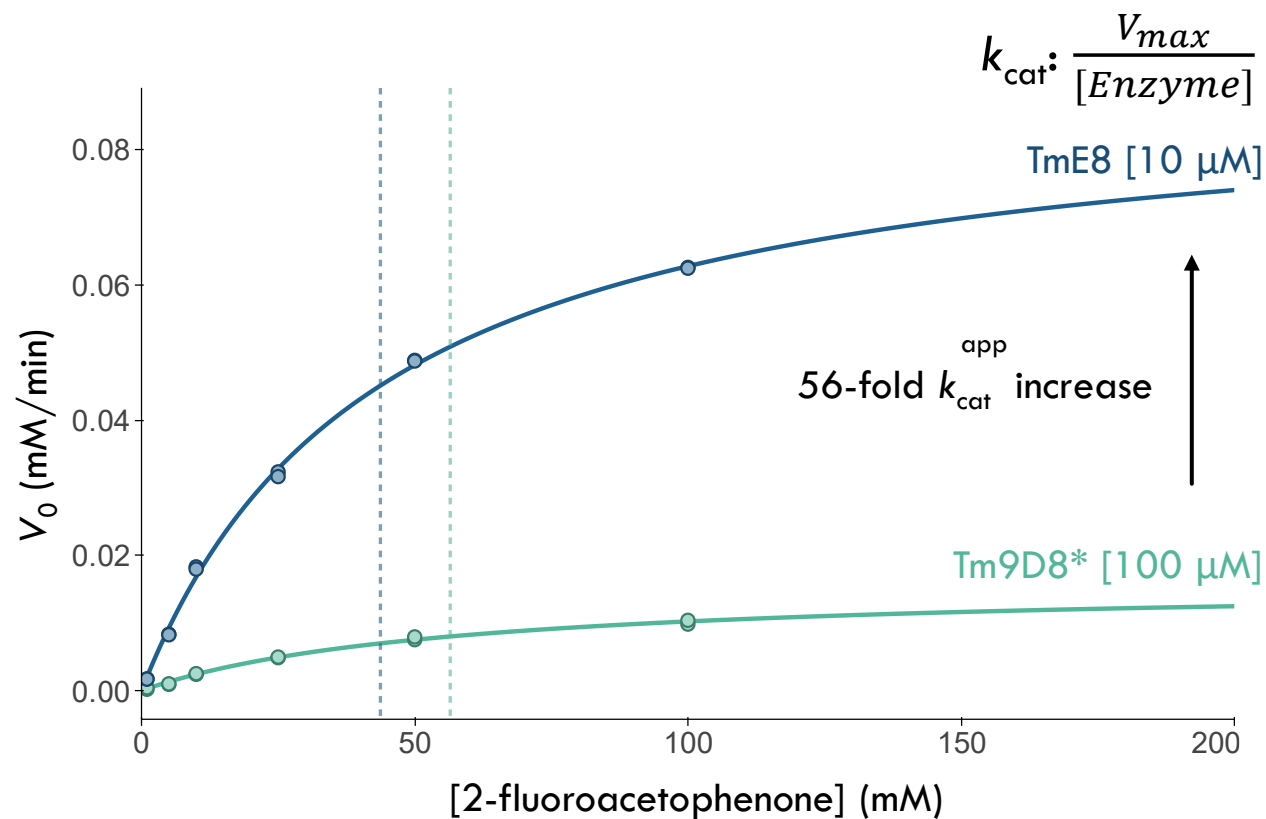
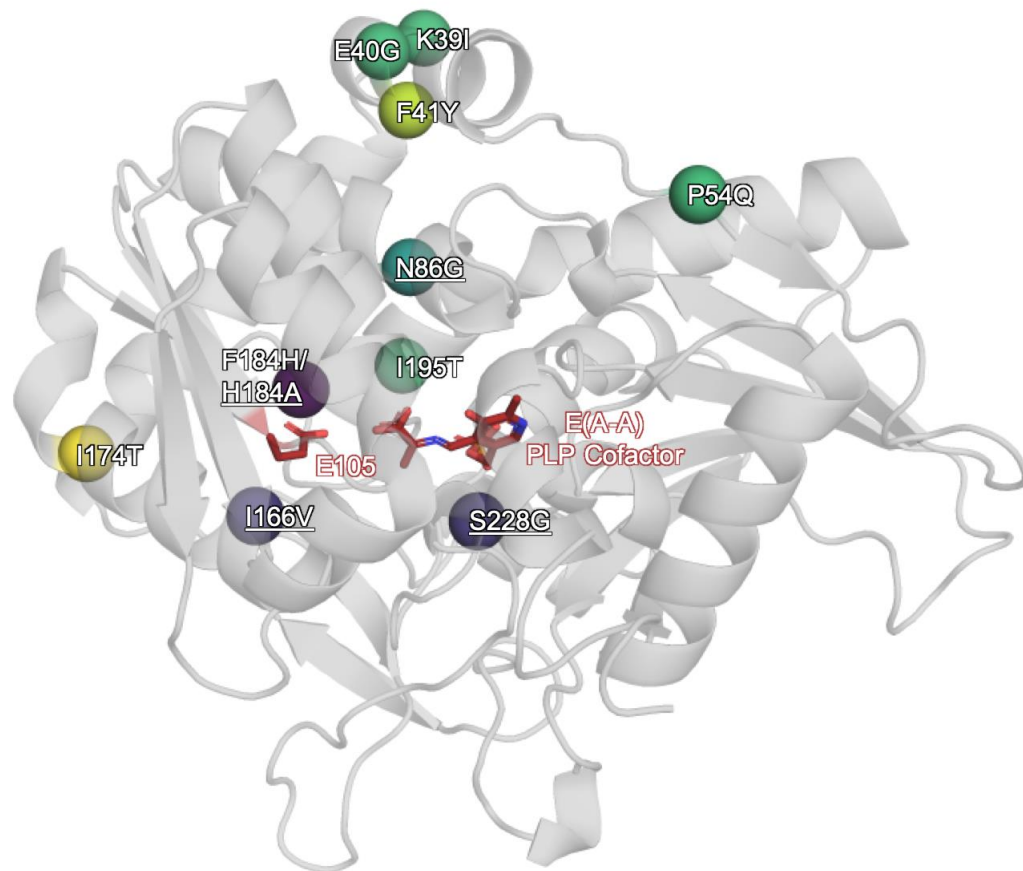
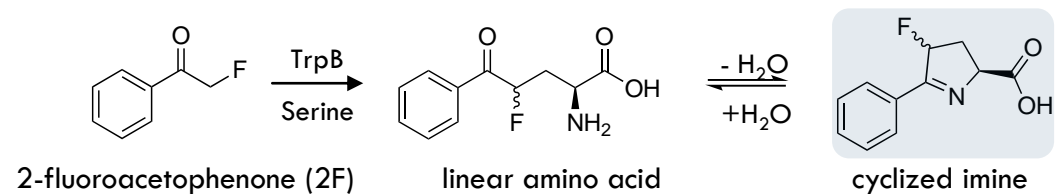


But if you try sometimes, well you might find, you get what you need!



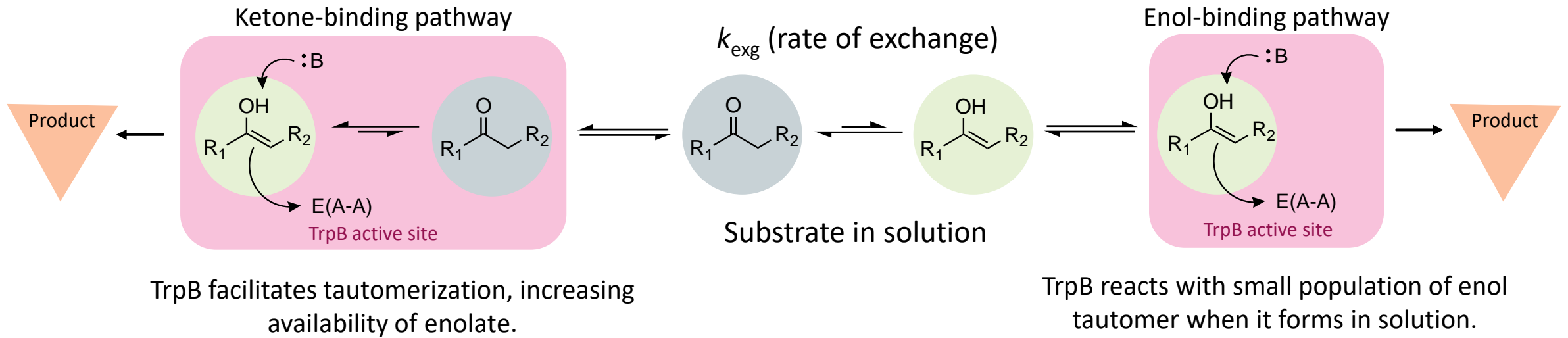
Selectivity was entirely inverted! TrpB can make both diastereomers.

Characterization of TmEnolate

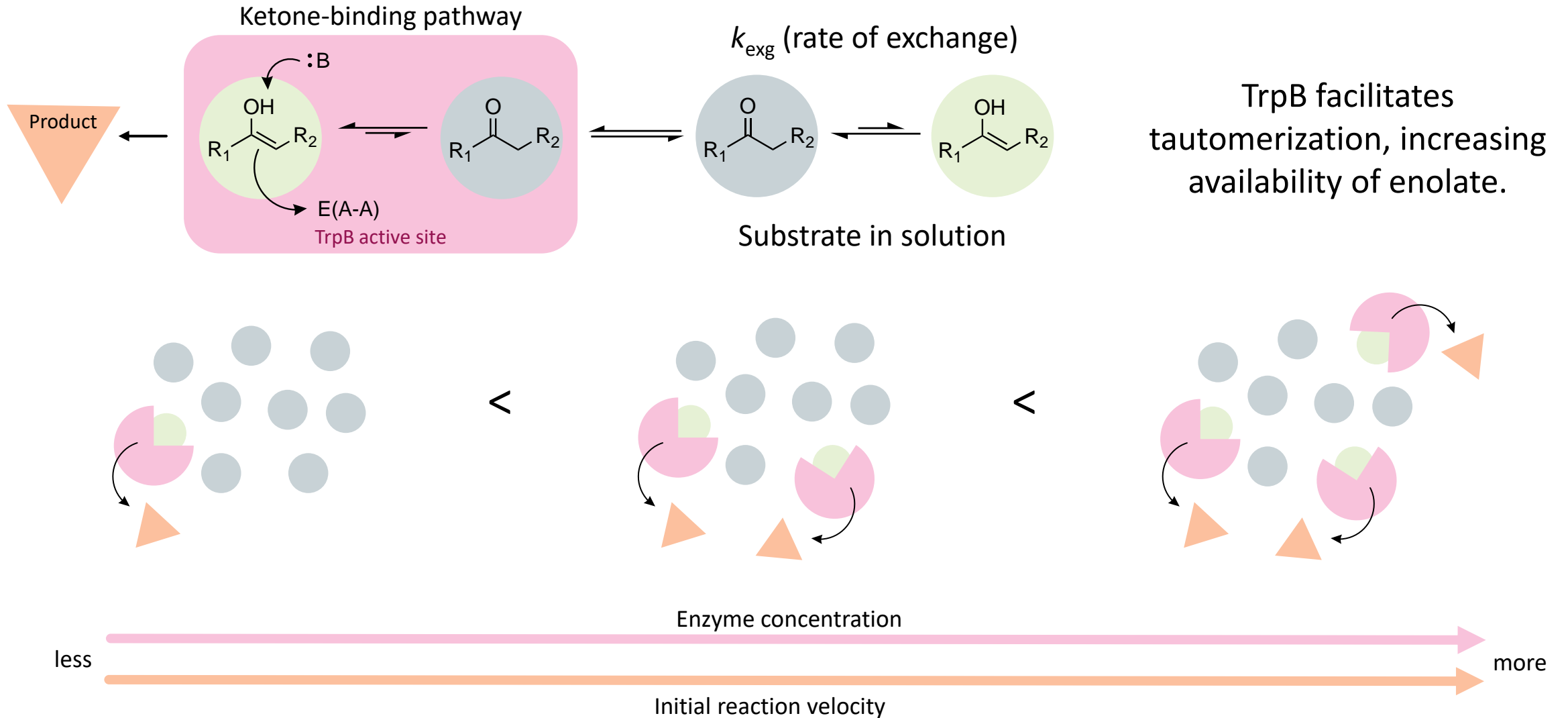


What did evolution do to speed up the reaction?

Does TrpB bind the ketone or enol?

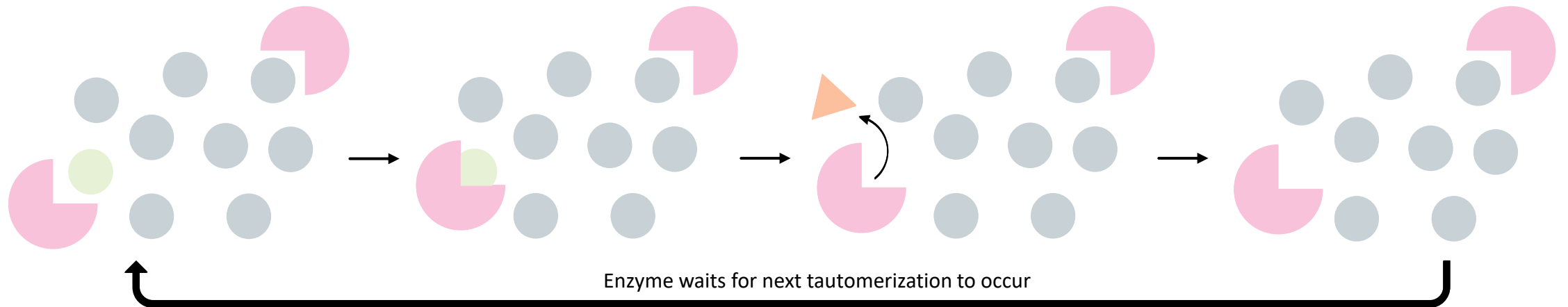
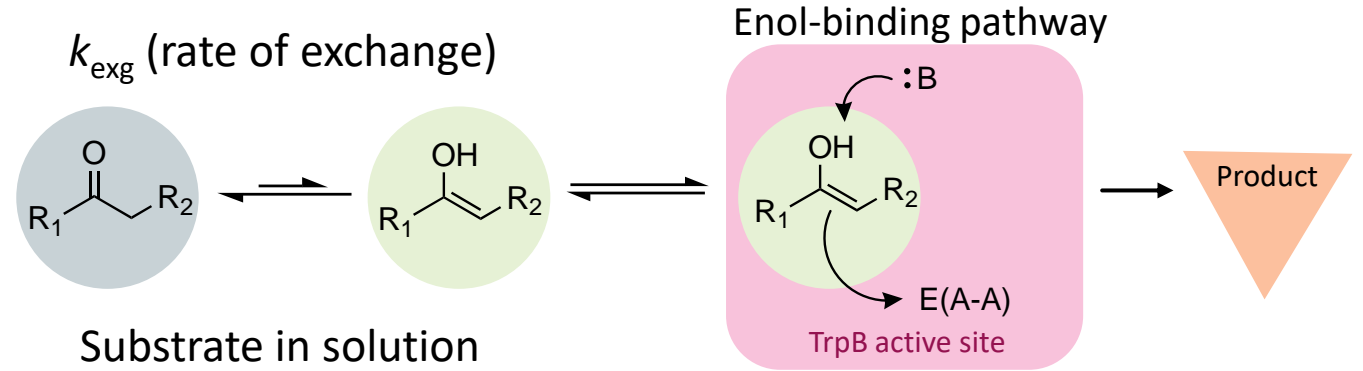


Does TrpB bind the ketone or enol?

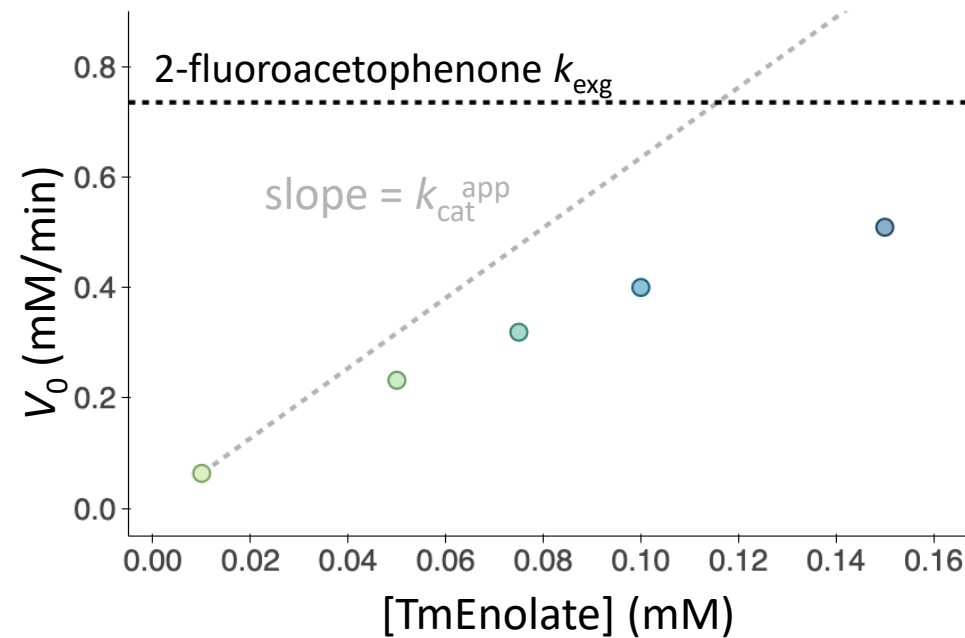
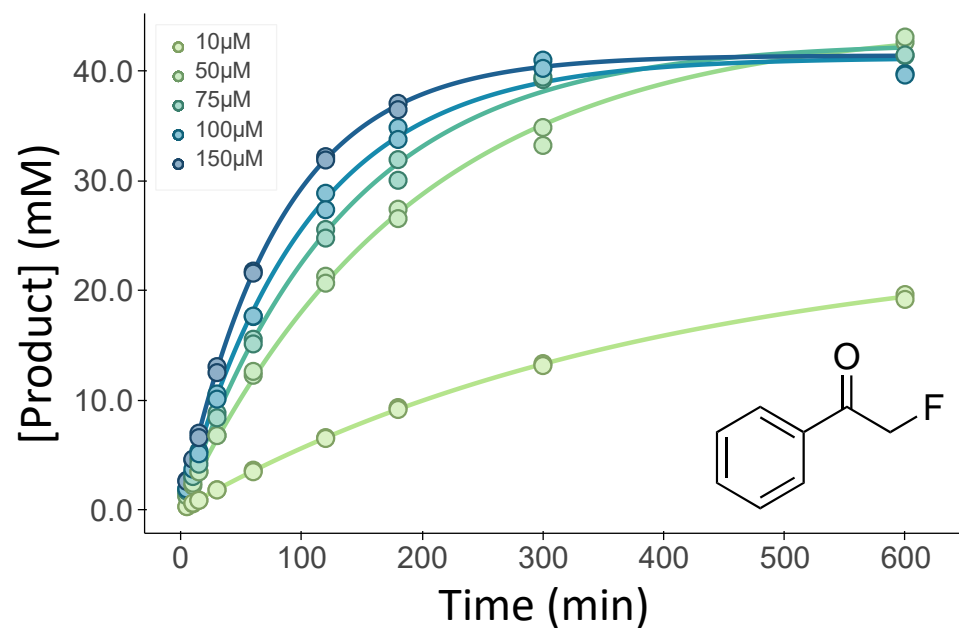
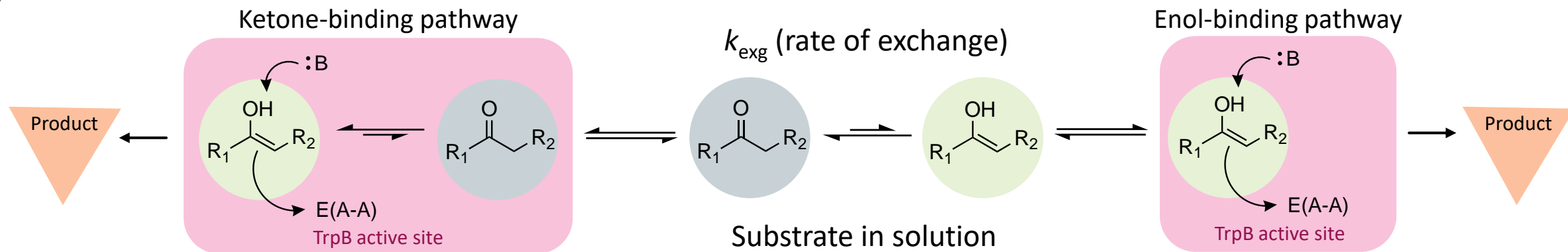


Does TrpB bind the ketone or enol?

TrpB reacts with small population of enol tautomer when it forms in solution



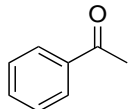
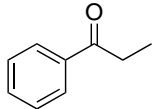
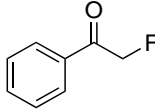
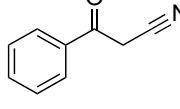
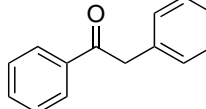
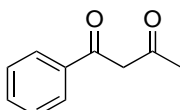
Does TrpB bind the ketone or enol?



[Enzyme] is not proportional to V_0 \therefore TrpB binds the enol

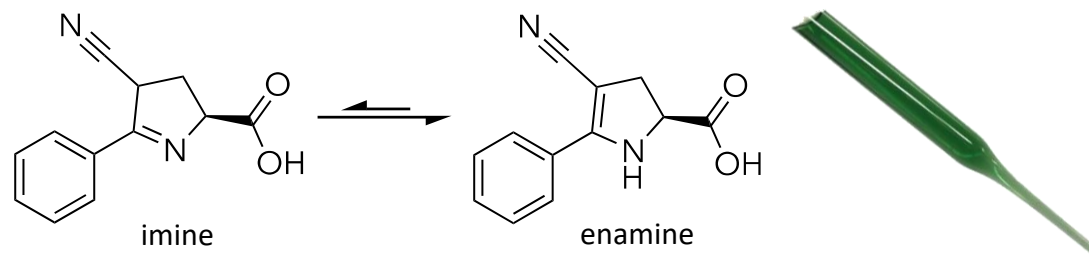
Product scope

k_{exg} is not the only determining factor for activity

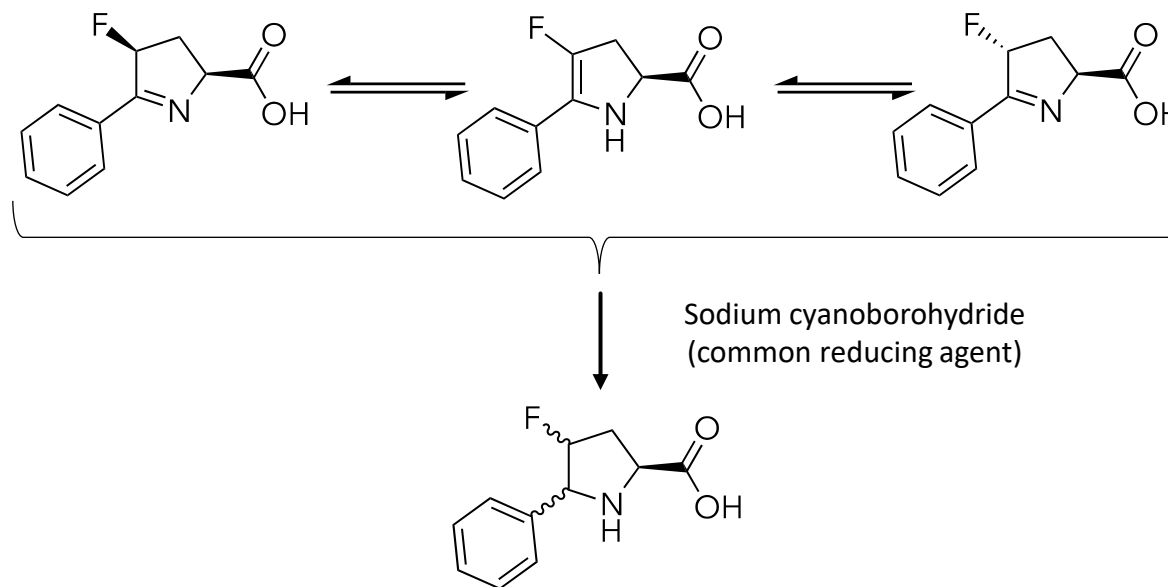
Entry	Substrate	TTN in 24 h	k_{exg} (min^{-1})
1			
2			
3			
4			
5			
6			

^[a] 2-CN-acetophenone is insoluble in water ^[b] Benzoylacetone hydrolyzes

Some products preferred the enamine state

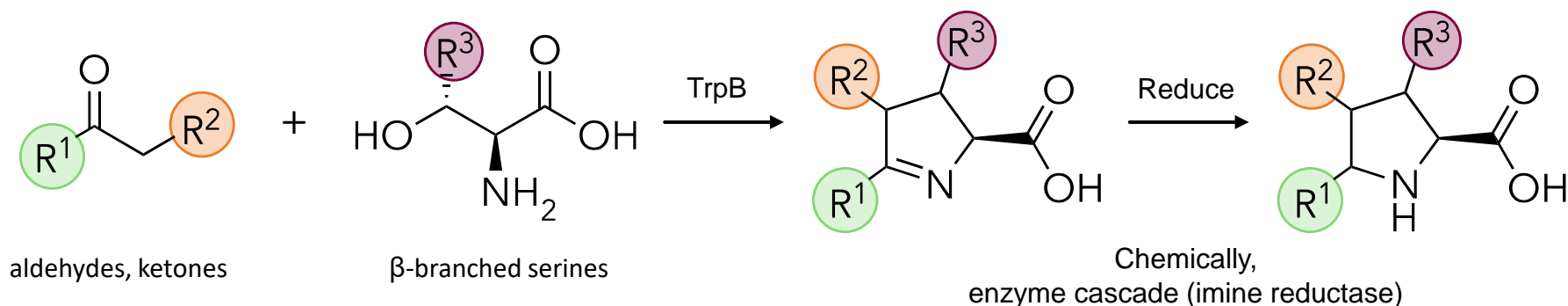


Reducing the product makes proline analogs!

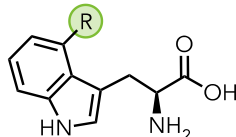
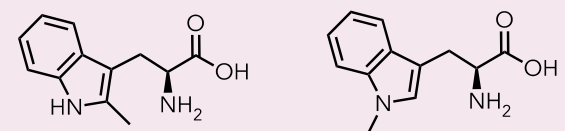
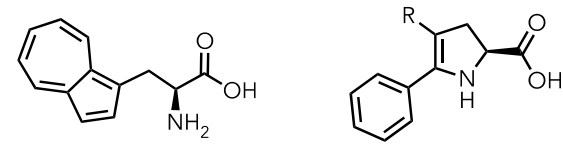
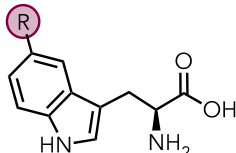
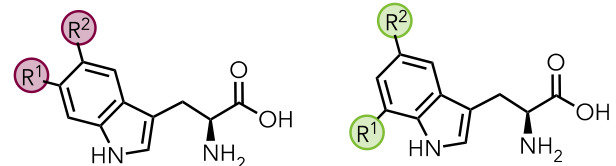
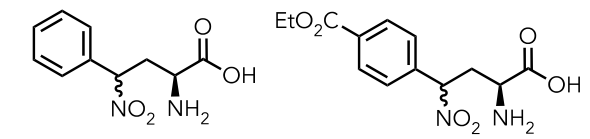
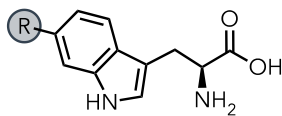
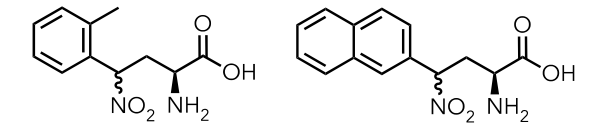
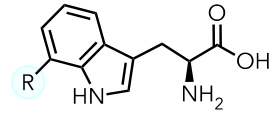
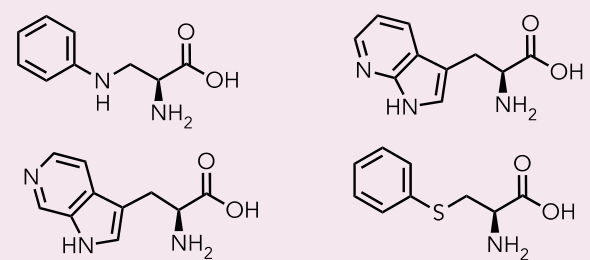
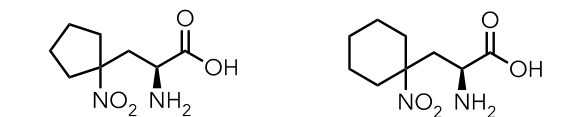
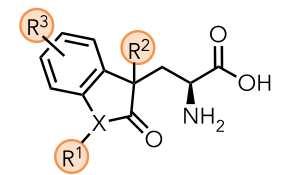
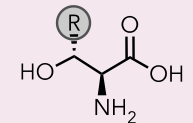


Summary and future directions

- Expanded the TrpBs nucleophile scope to include **ketones**.
- Demonstrated that **enzymes can intercept enolates**
 - No need for highly specific modes of activation
 - Kinetics, not thermodynamics, are what is important
 - Sets precedence for exploration of enolate chemistry in other enzymes
- In the future, proline analogs dense with chiral centers could be accessed using TrpB



New kids on the block

Substitution	EDG	Halogenated	EWG	1- and 2- substituted	Non-indole carbon nucleophiles
4	 CH ₃ OCH ₃ OH	F Cl Br	NO ₂ CO ₂ H CN CONH ₂		
5	 CH ₃ OCH ₃ OH	F Cl Br I	NO ₂ CHO CN CONH ₂ B(OH) ₂	Disubstituted  R ¹ , R ² = F, Cl, Br, I	
6	 CH ₃ OCH ₃ OH	F Cl Br	NO ₂ CN B(OH) ₂		
7	 CH ₃ OCH ₃ OH	F Cl Br I	NO ₂ CN	Other classes of α-amino acids: 	  R ¹ = CH ₃ , H , R ² = methyl, ethyl, propyl, butyl, OH, R ³ = Cl, FI
 β-substituted electrophiles accepted R = H, methyl, ethyl, propyl					

Buller AR, Brinkmann-Chen S, Romney DK, Herger M, Murciano-Calles J, Arnold FH *Proc Natl Acad Sci.* 2015;112(47):14599-14604.

Romney DK, Murciano-Calles J, Wehrmüller JE, Arnold FH. *J Am Chem Soc.* 2017;139(31):10769-10776.

Herger M, van Roye P, Romney DK, Brinkmann-Chen S, Buller AR, Arnold FH. *J Am Chem Soc.* 2016;jacs.6b04836.

Boville CE, Scheele RA, Koch P, Brinkmann-Chen S, Buller AR, Arnold FH. *Angew Chemie Int Ed.* 2018.

Romney, David K., Nicholas S. Sarai, and Frances H. Arnold. *ACS Catalysis* 2019.

Dick, Markus, et al., *J Am Chem Soc.* 141.50 (2019): 19817-19822.

Watkins, Ella J., Patrick J. Almhjell, and Frances H. Arnold. *ChemBioChem* 2019.

Watkins-Dulaney E.J., Dunham N.P., Straathof S., Turi S., Arnold F.H., Buller A.R.(2021) Manuscript in progress.

Acknowledgments

- Committee members
 - Mikhail Shapiro
 - Dave Tirrell
 - Rustem Ismagilov
 - Andrew Buller
- Team TrpB members
 - Andrew Buller
 - Tina Boville
 - David Romney
 - Patrick Almhjell
 - Sabine Straathof
- Enolates
 - Noah Dunham
 - Nicholas Porter
 - David Miller
 - Sabine Straathof
- General advice
 - Nat Goldberg
 - Anders Knight
- TrpB collaborators
 - Chang Liu
 - Gordon Rix

Frances Arnold



Sabine Brinkmann-Chen



Thank you to the entire Arnold group! You made my time at Caltech very special.

